

**NOVEL SYNTHESIS OF CYANOACETYLHYDRAZONE DERIVATIVES AND THEIR
ANTIMICROBIAL AND ANTIOXIDANT ACTIVITIES**

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**PERPUSTAKAAN
UNIVERSITI MALAYSIA SABAH**

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ABSTRACT

In this study, cyanoacetylhydrazone derivatives that possessing an azometine (-NHN=CH-) proton constitute an important class of compounds was synthesized and their biological activities were evaluated. The structures of newly synthesized compounds were confirmed by elemental analysis, FTIR spectral data. The range of the infrared region used is (4000~650 cm^{-1}). All synthesized compounds were tested and evaluated as antimicrobial and antiradical agents. In the determination of antimicrobial activity, the synthesized compounds showed no degree of inhibition against two types of bacterial namely *Streptococcus pneumonia* (G+) and *Escherichia coli* (G-) due to low percentage of purity of the synthesized compounds. A disk-diffusion method was use to evaluate the antimicrobial property of those compounds and Kirby-Bour disk-diffusion method was applied with modification. The antioxidant activity of synthesized compounds was determined by 1,1-Diphenyl-2-picrylhydrazyl (DPPH) assay. The synthesized compounds showed weaker antiradical activity (higher IC_{50}) than BHT which showed 0.0686 mg/mL. In conclusion, four types of cyanoacetylhydrazone compounds were synthesized and showed no inhibition in antimicrobial and lower potential in antioxidant properties. Hence, future study can be carried out to investigate and improve the effectiveness and purity of producing these compounds for better screening biological activities.

SINTESIS NOVEL TURUNAN CYANOACETYLHYDRAZONE DAN PENILAIAN AKTIVITI ANTIMIKROB DAN ANTIOKSIDAN

ABSTRAK

Dalam kajian ini, turunan cyanoacetylhydrazone memiliki sebuah proton azometine (-NHN=CH-) telah berjaya disediakan dan merupakan satu kelas yang penting untuk sebatian, berfungsi sebagai sasaran struktur dan aktiviti biologi telah dinilai. Struktur sebatian baru disintesis telah disahkan oleh analisis unsur, FTIR data spektrum. Rangkaian kawasan inframerah digunakan ialah dalam julat 4000 hingga 650 cm^{-1} . Sebatian wakil produk disintesis telah diuji dan dinilai sebagai agen antimikrob dan anti-radical. Dalam menentukan aktiviti antimikrob, sebatian yang telah disintesis tidak menunjukkan tahap perencatan terhadap dua jenis bakteria iaitu Streptococcus pneumonia (G+) dan Escherichia coli (G-) kerana peratusan kesucian sebatian disintesis yang rendah. Satu kaedah cakera penyebaran adalah digunakan untuk menilai harta antimikrob daripada sebatian dan pengubahsuaian Kirby-Bour kaedah cakera penyebaran turut digunakan. Aktiviti antioksidan sebatian disintesis telah ditentukan oleh 1,1-Diphenyl-2-picrylhydrazyl (DPPH) assay. Sebatian yang telah sintesis menunjukkan aktiviti antiradical yang lemah (lebih tinggi IC_{50}) daripada BHT yang menunjukkan 0.0686 mg/mL. Sebagai kesimpulan, empat jenis sebatian cyanoacetylhydrazone telah disintesis di mana perencatan dalam potensi anti-mikrob tidak ditunjukkan dan mempunyai potensi antioksidan yang rendah. Oleh itu, kajian pada masa hadapan dapat dijalankan untuk mengkaji dan meningkatkan keberkesanan dan ketulenan dalam menghasilkan sebatian tersebut untuk aktiviti biologi yang lebih baik.

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LIST OF SYMBOLS, UNITS AND ABBREVIATIONS

°C	Degree Celsius
%	Percentage
>	More than
&	And
wt	Weight
IR	Infrared
mol/L	Mol per liter
<	Less than
FTIR	Fourier Transform Infrared Spectroscopy
ml	Milileter
g	Gram
M	Molar
cm ⁻¹	Per centimeter
kg	Kilogram
g/mole	gram per mole
H ₂ O	Water
α	Alpha
β	Beta
Cl	Chlorine
Br	Bromine
NO ₂	Nitrogen Dioxide
C-N	Carbon-Nitrogen
N-N	Nitrogen-nitrogen
C=N	Carbon=Nitogen
-C=N-N-	Triatomic Group
KOH	Potassium Hydroxide
μg	Microgram
ml	Milliliter

CHAPTER 1

INTRODUCTION

1.1 Background of Study

The essence of organic synthesis is the formation of both single and multiple bonds between two or more carbon atoms or between many combinations of carbon. It also consist of various heteroatoms, most notably oxygen, nitrogen, sulfur and phosphorus, which may or may not be present in the final products of a particular synthetic scheme (Donohoe *et al.*, 2014; Voinov & Grigor'ev, 2002; Zhang *et al.*, 2014). Along the way to a particular target, such bond-forming reactions have to properly set up and this involves manipulations of many initial functional groups into other forms to enable the bond formation to take place or to provide ways around selectivity or group incompatibilities, either with each other or with the necessary reaction conditions (Knight, 2013).

Synthetic organic chemistry is one of the most rapidly developing fields in chemistry. Worldwide, useful new reagents and reactions are reported in the chemical literature daily. Among them, carbon-carbon double bond formation is one of the most useful and fundamental reactions in synthetic organic chemistry, particularly in the synthesis of complex naturally occurring products, which show biological activity (Shen, 2006). The resistance towards available drugs is rapidly evolving into a major threat to the global health security. To deal with this resistance, the need to design new compounds with better clinical efficacy has become one of the most important areas of today's research (Bala *et al.*, 2013).



1.1.1 Cyanoacetic Acid Hydrazide

Among several commercially available substituted hydrazines, cyanoacetic acid hydrazide has received the most attention recently. It is a versatile and convenient intermediate for the synthesis of various heterocyclic compounds. Figure 1.1 shows the chemical structure of cyanoacetic acid hydrazide (Martins *et al.*, 2008). The β -functional nitrile moiety of the molecule is a favorable unit for addition followed by cyclization or via cycloaddition with numerous reagents providing heterocyclic compounds of different ring sizes with one or several heteroatoms that are interesting as pharmaceuticals, as herbicides, as antibacterial agents, and as dyes. Their reactions with dinucleophiles normally result in the formation of polycyclic ring systems which may be the skeleton of important heterocyclic compounds. As reported in previous publications, β -functional nitriles had been used as starting components for novel synthesis of azoles, azines, and azoloazines. Among the β -functional nitriles, cyanoacetic acid hydrazide and their analogues are especially important starting materials or intermediates for the synthesis of various nitrogen-containing heterocyclic compounds (Bondock *et al.*, 2006).

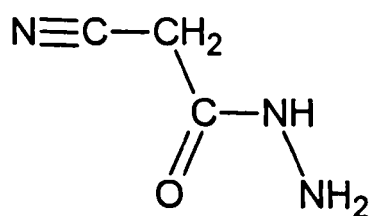


Figure 1.1 Structure of cyanoacetic acid hydrazide (Martins *et al.*, 2008).

1.1.2 Hydrazone

Hydrazones are a class of organic compounds which possess the structure $R_1R_2C=NNH_2$. They are related to ketone and aldehyde in which oxygen has been replaced with NNH_2 group. These azometine $-NHN=CH-$ proton constitute an important class of compounds for new drug development. Hydrazones are formed by the reaction of hydrazine or hydrazide with aldehydes and ketones. They act as reactants in various important reactions such as hydrazone iodination, Shapiro reaction and Bamford-Stevens reaction to form vinyl compounds. They act as intermediate in Wolff-Kishner reaction (Ali *et al.*, 2012).

Hydrazones contain two nitrogen atoms of different nature and a C-N double bond that is conjugated with a lone electron pair of the terminal nitrogen atom. These structural fragments are mostly responsible for the physical and chemical properties of hydrazones. The C-atom in hydrazone has both electrophilic and nucleophilic character and both the N-atoms are nucleophilic although the amino type nitrogen is more reactive. Due to these properties hydrazones are widely used in organic synthesis. Figure 1.2 shows the classification of active centers (Belskaya *et al.*, 2010).

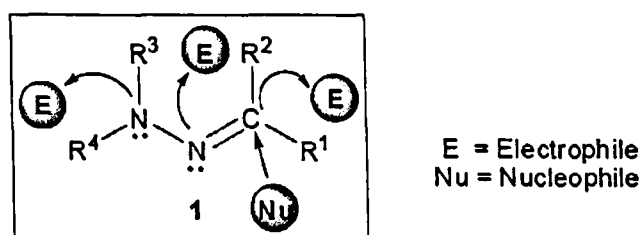
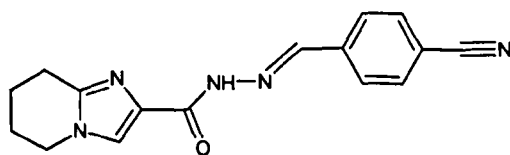
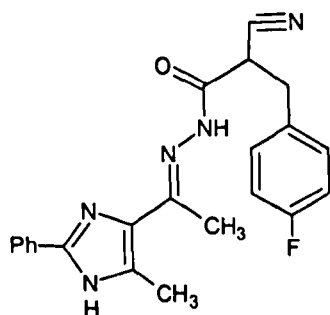


Figure 1.2 Classification of active centers (Belskaya *et al.*, 2010).

Hydrazones can also be synthesized by the Japp-Klingemann reaction (from β -ketoacids or β -ketoester's and aryldiazonium salts). The N,N'-dialkyl type of hydrazones can be hydrolysed, reduced and oxidized. This leads to the formation of amines by reduction of N-N bond. The C=N double bond in hydrazones are important compounds in drug design as they act as ligands for metal complexes, organocatalysis and synthesis of organic compounds. The chemical structure of hydrazone derivatives were showed in Figure 1.3 (Ali *et al.*, 2012).



N-[(*E*)-(4-cyanophenyl)methylidene]-5,6,7,8-tetrahydroimidazo[1,2-*a*]pyridine-2-carbohydrazide



2-cyano-3-(4-fluorophenyl)-*N*-[(1*E*)-1-(5-methyl-2-phenyl-1*H*-imidazol-4-yl)ethylidene]propanehydrazide

Figure 1.3 Hydrazones derivatives (Ali *et al.*, 2012).

Hydrazones are present in many of the bioactive heterocyclic compounds that are of wide interest due to their diverse biological and clinical applications. This created interest in researchers who have synthesized variety of hydrazone derivatives and screened them for their various biological activities viz. antimicrobial, anticonvulsant, antiviral, antidepressant, antimycobacterial, analgesic, anti-inflammatory, anthelmintic, antiplatelet, antimalarial, vasodilator, antischistosomiasis, anti-HIV, antidiabetic, anticancer and trypanocidal activities. Hydrazones possessing an azometine -NHN=CH- proton constitute an important class of compounds as target structures and evaluated their biological activities. These observations have been guiding for the development of new hydrazones that possess varied biological activities. Hydrazone-hydrazide compounds are not only intermediates but they are also very effective organic compounds in their own right. When they are used as intermediates, coupling products can be synthesized by using the active hydrozen component of -CONHN=CH- azometine group (Negi *et al.*, 2012).

In addition, some of the new hydrazone-hydrazones which are recently synthesized were active against *M. tuberculosis* H37Rv between the concentrations of 0.78-6.25 $\mu\text{g/mL}$. Synthesis of a series of hydrazone-hydrazones via the reaction of cyanoacetic acid hydrazide with bromo(4-methoxyacetophenone) reported for sedative, antidepressant and analgesic activities (Negi *et al.*, 2012). Development of

novel chemotherapeutic agents is an important and challenging task for the medicinal chemists and many research programs are directed towards the design and synthesis of new drugs for their chemotherapeutic usage. Hydrazone compounds constitute an important class for new drug development in order to discover an effective compound against multidrug resistant microbial infection (Deep *et al.*, 2010).

1.2 Objectives of Research

The objectives of this study are;

- i. to synthesize and characterize cyanoacetylhydrazone derivatives.
- ii. to determine the antibacterial and antioxidant activities of the synthesized cyanoacetylhydrazone derivatives.

1.3 Scope of Study

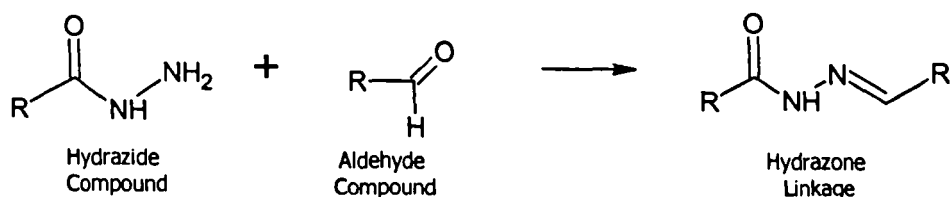
In this research, the cyanoacetylhydrazone is prepared by reacting acetophenone and cyanoacetic acid hydrazide. Cyanoacetylhydrazone are a very important group of analytical reagents for the determination of various biological assays. This has sparked a great interest for many researchers to synthesize vast varieties of hydrazone derivatives and screen for their various biological activities like anticonvulsant, antidepressant, analgesic, anti-inflammatory, antimicrobial, antioxidant and other biological activity. In this study, two entire new cyanoacetylhydrazone derivatives are synthesized and characterized. Biological test like antibacterial activity against Gram-negative and Gram-positive is assessed for these new synthesized products. Besides that, an antioxidant activity is also tested to develop the biological potential in pharmaceutical industry. In future, more cyanoacetylhydrazone derivatives can be further study and compare down to its atomic level to determine which functional group of the compound gives the main effect in the biological activity.

CHAPTER 2

LITERATURE REVIEW

2.1 Cyanoacetylhydrazone

Generally, Cyanoacetic acid hydrazide is used to synthesize hydrazone-hydrazone (Rafat M Mohareb *et al.*, 2011). Derivatives of hydrazine, particularly the hydrazide compounds formed from carboxylate groups, are able to react specifically with aldehyde or ketone functional groups in targeted molecules. A hydrazone linkage, which is a type of Schiff base, is created when reaction with either group takes place. Scheme 2.1 shows the reaction of hydrazide compound with aldehyde compound. A relatively stable bond is formed between the hydrazide compound and a ketone. However, the bond formed with an aldehyde group can be kinetically unstable (Hermanson, 2013).



Scheme 2.1 Formation of Hydrazone Linkage (Hermanson, 2013).

The reaction of cyanoacetic acid hydrazide with ω -bromo(4-methoxyacetophenone) gave the hydrazide-hydrazone derivative. These compounds reacted with either potassium cyanide or potassium thiocyanide to give the cyanide or thiocyanide derivatives respectively. The reaction of hydrazide-hydrazone

derivative with either hydrazine hydrate or phenyl hydrazine gave the hydrazine derivatives respectively. The latter compounds underwent a series of heterocyclization when react with different reagents to give 1,3,4-triazine and pyridinederivatives. These all the novel synthesized compounds show antidepressant, sedative and analgesic activities of the newly synthesized products were evaluated (Rafat & Mohamed, 2010).

Well known hydrazine derivatives, Cyanoacetic acid hydrazide or cyanoacetyl hydrazine, represents an important class of compounds that has greatly contributed in organic synthesis. Traditionally hydrazines have been employed as reagents for the derivatization and characterization of carbonyl compounds. However, in recent years the N-N linkage is found to be a key structural motif in various bioactive agents. Most significantly, an increase in number of N-N bond-containing heterocycles and peptidomimetics have emerged in commercial applications, such as pharmaceutical and agricultural agents (Mohareb et al., 2011). Research shows that Cyanoacetic acid hydrazide is a vital starting material when it reacts with pregnenolone to form hydrazide-hydrazone derivative followed by its heterocyclizations to form novel heterocyclic ring systems attached to pregnenone showed high inhibitory effects against non-small cell lung cancer (NCI-H460) and breast adenocarcinoma (MCF-7), which are much higher than the doxorubicin (Mohareb & Al-Omran, 2012). Cyanoacetic acid hydrazide also act as starting material, which reacts with cycloalkanones to form cyanoacetohydrazones which is useful in the synthesis of 6-Amino-2-oxo-3,5-pyridinedicarbonitriles that possesses biological properties like antihypertensive, antihistaminic, anticancer, in addition to antibacterial or antifungal (Girgis *et al.*, 2003).

Cyanoacetic acid hydrazide is commercially available (Martins *et al.*, 2008). Cyanoacetic acid hydrazide, together with its derivatives, are important groups of starting material able to produce various types of compounds which have been reported to possess varied biological and pharmacological activity. Many publications and research has reported synthesis of different heterocyclic compounds, which utilizes cyanoacetic acid hydrazide as the key starting material. Some of the biological properties of heterocyclic compounds that were prepared from cyanoacetic acid hydrazide were reported (Abubshait, 2012). Aziz & Gomha had successfully

synthesized different series of 1,4-dihydropyridine derivatives by using cyanoaceto-hydrazone with different aromatic and heteromatic ketones as starting material. Screened for cytotoxic effects of the 1,4-dihydropyridine derivatives against human breast cell line MCF-7 had been done using Sulfo-Rhodamine-B stain (SRB) assay method (Aziz & Gomha, 2013). Eldin *et al.*, used cyanoacetic acid hydrazone in heterocyclic synthesis of a few annelated pyran derivatives, which shows latent functional substituents, making them highly promising for biological activity studies as well as for further chemical transformations (Eldin *et al.*, 1993). Doss *et al.*, reacted cyanoacetic acid hydrazone with Digitoxin and Digoxin to produce the hydrazone-hydrazone derivatives, which is helpful when synthesizing various type of coumarin, thiazole, thiophene and pyridine derivatives with potential biological activities (Doss *et al.*, 2001).

Cyanoacetic acid hydrazone can be used as an ambident nucleophile, which means that it can have properties of both *N*- and *C*- nucleophile. During treatment of cyanoacetic acid hydrazone with various reagents, there are five possible sites for the reaction or attack to take place; the nucleophile is able to attack the carbon of the carbonyl function (position 3) and the carbon atom of the nitrile function (position 5). On the other hand, the active methylene group (position 4) and amino groups (positions 1 and 2) are able to attack electrophiles. Figure 2.1 illustrates the five possible sites of cyanoacetic acid hydrazone (Bondock *et al.*, 2006).

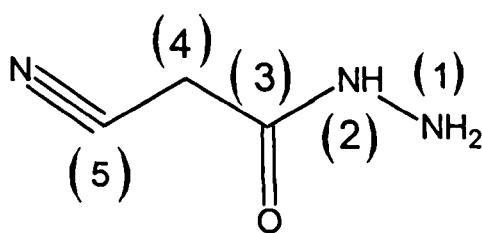
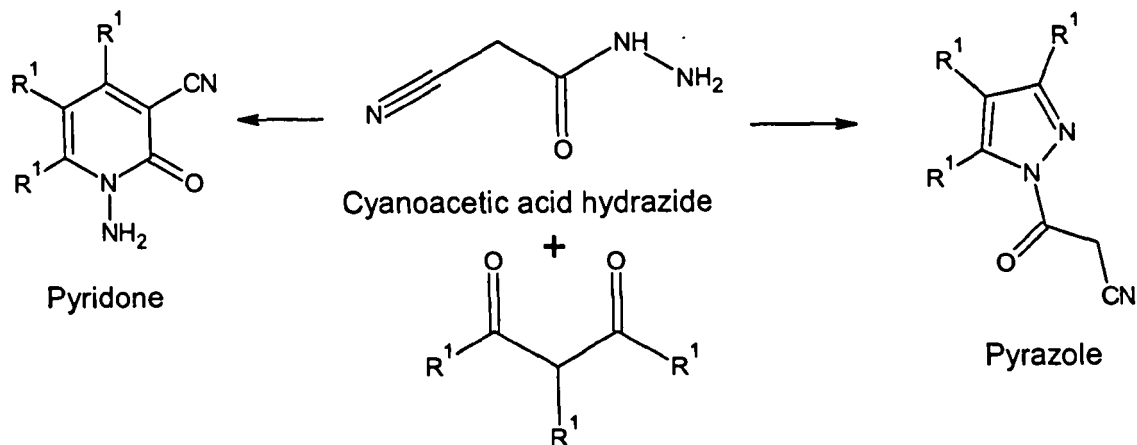


Figure 2.1 Cyanoacetic Acid Hydrazone (Bondock *et al.*, 2006).

The reactions of cyanoacetic acid hydrazone with wide number of reactants, both nucleophiles and electrophiles, are used to synthesize a variety of polyfunctional heterocyclic compounds with potential biological interest. Reports have shown that the main reaction involves the cyclocondensation reaction of cyanoacetic acid hydrazone with 1,3-dicarbonyl compounds. These reports have shown prove, where

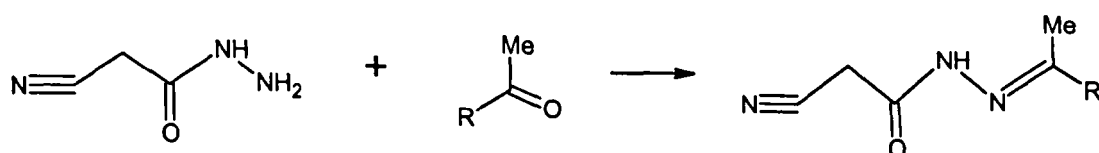
the product of this cyclocondensation reaction is dependent on the reaction conditions. Scheme 2.2 demonstrates the reaction of cyanoacetic acid hydrazide (Martins *et al.*, 2008).



Scheme 2.2 Reaction of Cyanoacetic acid hydrazide (Martins *et al.*, 2008).

2.2 Cyanoacetylhydrazone Derivatives

Cyanoacetohydrazones is said to possess interesting antimicrobial, antitumor, antihepatitis-C virus (HCV activity). Treatment of cyanoacetohydrazide with acetophenone derivatives in ethanol, with presence of catalytic amount of acetic acid gives 2-cyano-N-(1-substituted ethylidene) acetohydrazides respectively in good yields. Scheme 2.3 shows the formation of cyanoacetohydrazone. Reaction of cyanoacetohydrazide with different aromatic and heteroaromatic ketones yielded the corresponding hydrazones, which when reacted with ketene dithioacetal in the presence of KOH at room temperature, gives pyridinedicarbonitrile derivatives. The latter compounds, when refluxed with hydrazine hydrate, gives pyrazolo[4,3-*c*]pyridinecarbonitrile derivatives. The potential of these derivatives were then evaluated for their *in-vitro* cytotoxic activity against human breast cancer cell line (MCF7). Some of the compounds revealed significant activity compared to Doxorubicin as a reference drug (Aziz & Gomha, 2013).



Scheme 2.3 Formation of Cyanoacetohydrazone (Aziz & Gomha, 2013).

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