INVESTIGATION ON THE GOLD NANOPARTICLES CONCENTRATION EFFECTS TO THE OPTICAL AND STRUCTURAL PROPERTIES OF NANOSTRUCTURE GLASS

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JUDUL: **INVESTIGATION ON THE GOLD NANOPARTICLES CONCENTRATION EFFECTS TO THE OPTICAL AND STRUCTURAL PROPERTIES OF NANOSTRUCTURE GLASS**

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DECLARATION

I hereby declare that the work in this thesis is my own except for quotation, equations, summaries and references which have been acknowledged.

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ABSTRACT

Achieving enhancement of photoluminescence and Raman spectra of glasses containing rare-earth ions and metallic nanoparticles are prerequisite. Many studies only reported the effect of concentration of the nanoparticles on structural and optical properties of the glasses instead of the variation in size and shape of nanoparticles. Thus, the current study aims to investigate effect of concentration of the gold (Au) nanoparticles on structural and optical properties of the glasses. Further, their geometric effect on photoluminescence and Raman enhancement with varying concentration of the gold nanoparticles for optimizing photoluminescence and Raman enhancement are addressed. A series of glass with composition of $70TeO_2$-$20ZnO$-$10Na_2O$-$0.5Er_2O_3$-(x)Au where x=0.0, 0.1, 0.2, 0.3 and 0.4 mol% were examined with X-ray diffraction (XRD), ultraviolet-visible (UV-Vis) spectroscopy, transmission electron microscopy (TEM), scanning electron microscopy (SEM), energy dispersive X-ray spectroscopy (EDX), photoluminescence spectroscopy, Fourier Transform Infrared (FTIR) spectroscopy and Raman spectroscopy. Modification in UV-Vis spectra was observed due to non-bridging oxygen created by gold nanoparticles with increasing concentration of gold nanoparticles. TEM images verified the presence of non-spherical gold nanoparticles. Photoluminescence spectra revealed luminescence enhancement caused by the effect of surface plasmon resonance (SPR) and energy transfer from the gold nanoparticles to erbium ions as concentration of the gold nanoparticles was increased up to 0.3 mol%. However, when the concentration of the gold nanoparticles exceeded 0.3 mol%, luminescence quenching occurs due to energy transfer from erbium ions to gold nanoparticles. Amplification of Raman spectra was attributed to the effect of SPR. The concentrations of the gold nanoparticles for maximizing photoluminescence and Raman spectra were found to be 0.3 and 0.4 mol%, respectively. Non-spherical shaped nanoparticles were found to optimize photoluminescence and Raman enhancement. The results demonstrated the effect of varying concentration of nanoparticles on the properties of the glasses due to the geometric effect of the nanoparticles especially on the enhancement of photoluminescence and Raman spectra.

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ABSTRAK

KAJIAN KESAN KEPEKATAN ZARAH NANO EMAS TERHADAP SIFAT OPTIK DAN STRUKTUR BAGI KACA BERSTRUKTUR NANO

Peningkatan spektra fotoluminesen dan Raman bagi kaca yang mengandungi ion-ion bumi nadir dan zarah nano logam adalah sangat penting. Kebanyakan kajian hanya melaporkan pengaruh kepekatan zarah nano terhadap sifat-sifat struktur dan optik kaca tersebut tetapi tidak melaporkan kepelbagaian pada saiz dan bentuk zarah nano. Maka, kajian ini bertujuan untuk mengkaji kesan kepekatan zarah nano emas pada sifat struktur dan optik kaca. Tambahan pula, kesan geometri pada peningkatan fotoluminesen dan Raman dengan kepelbagaian zarah nano emas untuk mendapatkan peningkatan fotoluminesen dan Raman yang optimum dibincangkan. Satu siri kaca yang mempunyai komposisi $70TeO_2$-$20ZnO$-$10Na_2O$-$0.5Er_2O_3$-$(x)Au$ di mana x=0.0, 0.1, 0.2, 0.3 dan 0.4 mol% telah dikaji menggunakan teknik pembelauan sinar-X (XRD), spektroskopi ultraungu-cahaya nampak (UV-Vis), mikroskopi pancaran elektron (TEM), mikroskopi pengimbasan elektron (SEM), spektroskopi serakan tenaga sinar-X (EDX), spektroskopi fotoluminesen, spektroskopi Inframerah Pengubah Fourier (FTIR) dan spektroskopi Raman. Perubahan pada spektrum UV-Vis disebabkan oleh ikatan oksigen tidak bersambung yang dihasilkan oleh zarah nano emas apabila kepekatan zarah nano emas ditingkatkan. Imej TEM membuktikan kehadiran zarah nano emas yang mempunyai bentuk bukan sfera. Spektrum fotoluminesen menunjukkan peningkatan daripada kesan resonan plasmon permukaan (SPR) dan pemindahan tenaga daripada zarah nano emas kepada ion-ion erbium apabila kepekatan zarah nano emas ditingkatkan sehingga 0.3 mol%. Bagaimanapun, apabila kepekatan zarah nano emas melebihi 0.3 mol%, penurunan pada keamatan fotoluminesen berlaku disebabkan pemindahan tenaga daripada ion-ion erbium kepada zarah nano emas. Peningkatan spektrum Raman disebabkan oleh kesan SPR. Didapati kepekatan zarah nano emas untuk mengoptimumkan peningkatan fotoluminesen dan Raman ialah 0.3 dan 0.4 mol%. Zarah nano berbentuk bukan sfera mengoptimumkan peningkatan fotoluminesen dan Raman. Keputusan tersebut menunjukkan kesan kepekatan zarah nano terhadap sifat-sifat kaca disebabkan kesan geometri terutamanya pada peningkatan spektra fotoluminesen dan Raman.

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LIST OF ABBREVIATIONS AND SYMBOLS

TeO_2	-	Tellurium dioxide
Na_2O	-	Sodium oxide
ZnO	-	Zinc oxide
Er_2O_3	-	Erbium (III) oxide
Au	-	Gold element
Te	-	Tellurium element
Zn	-	Zinc element
Na	-	Sodium element
Er	-	Erbium
O	-	Oxygen element
C	-	Carbon element
SPR	-	Surface plasmon resonance
XRD	-	X-ray diffraction
n	-	Order of diffraction based on Bragg's law of XRD
d	-	Spacing between two successive lattice planes of a crystal based on Bragg's law of XRD
θ	-	Angle of incidence of X-ray based on Bragg's law of XRD
λ	-	Wavelength of an electromagnetic wave (for the case of spectroscopy) or an electron (for the case of de Broglie equation for operation of TEM) or a characteristic X-ray emitted by an atom (for the case of analysis of EDX)
2θ	-	Angle between an X-ray source and an X-ray detector
β	-	Full-width at half-maximum (FWHM) of XRD peak due to

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crystallite based on Debye-Scherrer equation

Å	-	Angstrom
Cu	-	Copper
Kα	-	K alpha of X-ray emitted by copper atoms in copper anode in XRD of current study
$K\alpha_1$	-	K alpha 1 (One of K alpha X-ray doublets emitted by copper atoms in copper anode in XRD of current study)
$K\alpha_2$	-	K alpha 2 (One of K alpha X-ray doublets emitted by copper atoms in copper anode in XRD of current study)
nm	-	Nanometer
UV-Vis	-	Ultraviolet-visible obtained from ultraviolet-visible spectroscopy
A	-	Absorbance of a glass sample
T	-	Transmittance of a glass sample given by $T = I/I_0$
a.u.	-	Arbitrary unit (a unit used by absorbance for UV-Vis spectra, intensity of excitation and emission spectra of photoluminescence spectroscopy, and intensity of Raman spectra)
α	-	Absorption coefficient
d	-	Thickness of a glass sample
I_0	-	Intensity of incident light on a glass sample
I	-	Intensity of transmitted light through a glass sample
dI	-	The differential of intensity of light to represent the change of intensity of light as the light passes through the glass sample
dx	-	The differential of distance travelled by light inside the glass sample to represent the amount of distance travelled by the light inside the glass sample

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h	-	Planck's constant equal to 6.626 x 10^{-34} J s where J s is pronounced as Joule second
ΔL	-	Change in orbital angular momentum L during a transition between two energy levels
ΔS	-	Change in spin angular momentum S during a transition between two energy levels
ΔJ	-	Change in total angular momentum J during a transition between two energy levels where $J = L + S$
eV	-	Electron volt (a unit used to quantify optical band gaps of glass samples and Urbach energy for UV-Vis spectral analysis)
$\hbar$	-	Planck's constant h divided by 2π
c	-	Speed of light in vacuum equal to 3.0 x 10^{8} m/s
ω	-	Angular frequency of light, $\omega = 2\pi f$ with f representing frequency of light
E_{opt}	-	Optical band gap
n	-	Exponent of a factor $(\hbar\omega - E_{opt})$ that depends on type of optical transition based on Davis-Mott theory
E_{dir}	-	Direct optical band gap
E_{indir}	-	Indirect optical band gap
E_U	-	Urbach energy
B	-	Energy-independent constant in mathematical expression of Davis-Mott theory
cm^{-1}	-	Per unit centimeter (a unit of energy for energy level diagram in section of photoluminescence spectroscopy; a unit of wavenumber for FTIR, and a unit of Raman shift for Raman spectroscopy)
ATR	-	Attenuated total reflection

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FTIR	-	Fourier Transform Infrared spectroscopy
n_i	-	Refractive index of a first medium that an infrared light is incident on based on principle of ATR-FTIR
n_r	-	Refractive index of a second medium that an infrared light passes through based on principle of ATR-FTIR
θ_i	-	Angle of incidence of an infrared light based on principle of ATR-FTIR
θ_r	-	Angle of refraction of an infrared light based on principle of ATR-FTIR
TEM	-	Transmission electron microscopy
p	-	Momentum of an electron based on de Broglie's equation in physical principle of TEM
U	-	Electrostatic potential difference (or accelerating voltage) to accelerate electrons emitted by electron guns of TEM
μ	-	Location parameter of a lognormal distribution function
σ	-	Shape parameter of a lognormal distribution function
SEM	-	Scanning electron microscopy
EDX	-	Energy dispersive X-ray spectroscopy; an abbreviation with the same meaning as that of EDS
E_i	-	Energy of an initial state of an electron of a given atom based on physical principle of EDX
E_f	-	Energy of a final state that has lower energy than that of an initial state of an electron of a given atoms based on physical principle of EDX
E_0	-	Beam energy of an incident electron beam in SEM and EDX
keV	-	Kilo electron volt (used in EDX spectrum in the current study)

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REFERENCES

Agilent Technologies, Inc. 2011a. *Agilent Technologies Cary 60 UV-Vis spectrophotometer: Guaranteed Specifications*. Santa Clara, CA: Agilent Technologies. Retrieved from https://www.agilent.com/cs/library/specifications/public/5990-7881EN_Cary60_Specifications.pdf on 20 April 2019.

Agilent Technologies, Inc. 2011b. *Agilent Technologies Cary 60 UV-Vis Spectrophotometer Brochure*. Santa Clara, CA: Agilent Technologies. Retrieved from https://www.agilent.com/cs/library/brochures/5990-7789EN_Cary_60_UV-Vis_Brochure.pdf on 20 April 2019.

Agilent Technologies, Inc. 2011c. *Agilent Technologies Cary 60 UV-Vis Spectrophotometer: Measuring Baseline-corrected Spectra on a Cary 60 UV-Vis*. Santa Clara, CA: Agilent Technologies. Retrieved from https://www.agilent.com/cs/library/technicaloverviews/public/5990-7947EN.pdf on 20 April 2019.

Agilent Technologies, Inc. 2011d. *Agilent Technologies Cary Eclipse Fluorescence Spectrophotometer: Guaranteed Specifications*. Santa Clara, CA: Agilent Technologies.

Agilent Technologies, Inc. 2011e. *Agilent Technologies Cary Eclipse Fluorescence Spectrophotometer: User's Guide*. Santa Clara, CA: Agilent Technologies. Retrieved from https://www.agilent.com/cs/library/usermanuals/public/1758.pdf on 20 April 2019.

Agilent Technologies, Inc. 2011f. *Agilent Technologies Cary Eclipse Fluorescence Spectrophotometer Brochure*. Santa Clara, CA: Agilent Technologies. Retrieved from https://www.agilent.com/cs/library/brochures/5990-7788EN_Cary_Eclipse_Brochure.pdf on 20 April 2019.

Ahmadi, F., Hussin, R. & Ghoshal, S.K. 2017. Spectroscopic Attributes of Sm^{3+} Doped Magnesium Zinc Sulfophosphate Glass: Effect of Silver Nanoparticles Inclusion. *Optical Materials*, 73, 268-276.

Akbari, B., Tavandashti, M. P. & Zandrahimi, M. 2011. Particle Size Characterization of Nanoparticles – A Practical Approach. *Iranian Journal of Materials Science & Engineering*, 8(2), 48-56.

Akbarian, F., Dunn, B. S. & Zink, J. I. 1996. Porous Sol-gel Silicates Containing Gold Particles as Matrices for Surface-enhanced Raman Spectroscopy. *Journal of Raman spectroscopy*, 27(10), 775-783.

UMS
UNIVERSITI MALAYSIA SABAH

Annapoorani, K., Murthy, N. S., Ravindran, T.R. & Marimuthu, K. 2016. Influence of Er^{3+} ion Concentration on Spectroscopic Properties and Luminescence Behavior in Er^{3+} Doped Strontium Telluroborate Glasses. *Journal of Luminescence*, 171, 19-26.

Arcidiacono, S., Bieri, N. R., Poulikakos, D. & Grigoropoulos, C. P. 2004. On the Coalescence of Gold Nanoparticles. *International Journal of Mulitphase Flow*, 30(7-8), 979-994.

ASEQ Instruments. no yr. *Rm1 – Raman spectrometer*. Vancouver, Canada: ASEQ Instruments. Retrieved from http://www.aseq-instruments.com/ASEQ_Raman.htm on 6 June 2018.

Awang, A., Ghoshal, S.K., Sahar, M.R., Dousti, M.R., Amjad, R. J. & Nawaz, F. 2013. Enhanced Spectroscopic Properties and Judd-Ofelt Parameters of Er-doped Tellurite Glass: Effect of Gold Nanoparticles. *Current Applied Physics*, 13(8), 1813-1818.

Awang, A., Ghoshal, S.K., Sahar, M.R., Dousti, M. R. & Nawaz, F. 2014. Growth of Au Nanoparticles Stimulate Spectroscopic Properties of Er^{3+}-doped TeO_2-ZnO-Na_2O Glasses. *Advanced Materials Research*, 895, 254-259.

Awang, A., Ghoshal, S.K., Sahar, M.R. & Arifin, R. 2016. Tailoring Spectroscopic Properties of Er^{3+} Doped Zinc Sodium Tellurite Glass via Gold Nanoparticles. *Jurnal Teknologi*, 78(3-2), 139-144.

Ayuni, J. N., Halimah, M. K., Talib, Z. A., Sidek, H. A. A., Daud, W. M., Zaidan, A. W. & Khamirul, A. M. 2011. Optical Properties of Ternary TeO_2-B_2O_3-ZnO Glass System. *IOP Conference Series: Materials Science and Engineering*, 17(1), 012027.

Babu, S. S., Jang, K., Cho, E. J., Lee, H. & Jayasankar, C. K. 2007. Thermal, Structural and Optical Properties of Eu^{3+}-doped Zinc-tellurite Glasses. *Journal of Physics D: Applied Physics*, 40(18), 5767-5774.

Baia, L., Baia, M., Kiefer, W., Popp, J. & Simon, S. 2006. Structural and Morphological Properties of Silver Nanoparticles-phosphate Glass Composites. Chemical Physics, 327(1), 63-69.

Baia, L. & Simon, S. 2007. UV-Vis and TEM Assessment of Morphological Features of Silver Nanoparticles from Phosphate Glass Matrices. *In* A. Méndez-Vilas & J. Díaz (eds.), *Modern Research and Educational Topics in Microscopy*, pp. 576-583. Badajoz, Spain: Formatex. Retrieved from http://citeseerx.ist.psu.edu/viewdoc/download?doi=10.1.1.605.2338&rep=rep1&type=pdf on 15 August 2018.

Ballarin, B., Cassani, M. C., Tonelli, D., Boanini, E., Albonetti, S., Blosi, M. & Gazzano, M. 2010. Gold Nanoparticles-containing Membranes from In Situ

UMS
UNIVERSITI MALAYSIA SABAH

Reduction of a Gold(III)-aminoethylimidazolium Aurate Salt. *The Journal of Physical Chemistry C*, 114(21), 9693-9701.

Bearden, J. A. 1967. X-ray Wavelengths. *Reviews of Modern Physics*, 39(1), 78-124.

Bhattacharya, S. & Ghosh, A. 2010. Relaxation Dynamics in Superionic Molybdate Glass Nanocomposites Embedded with α-AgI Nanoparticles. *The Journal of Physical Chemistry C*, 114(13), 5745-5750.

Bowen, P. 2007. Particle Size Distribution Measurement from Millimetres to Nanometers and from Rods to Platelets. *Journal of Dispersion Science and Technology*, 23(5), 631-662.

Brance, U. R., Drummond, I. W., Finbow, D., Harden, E. H. & Kenway, P. 1978. 27. Hydrocarbon Contamination in Vacuum Dependent Scientific Instruments. *Vacuum*, 28(10-11) 489-496.

Carnall, W. T., Fields, P. R. & Rajnak, K. 1968. Electronic Energy Levels in the Trivalent Lanthanide Aquo Ions. I. Pr^{3+}, Nd^{3+}, Pm^{3+}, Sm^{3+}, Dy^{3+}, Ho^{3+}, Er^{3+}, and Tm^{3+}. *The Journal of Chemical Physics*, 49(10), 4424-4442.

Çelikbilek, M., Ersundu, A.E., Solak, N. & Aydin, S. 2011. Crystallization Kinetics of the Tungsten-tellurite Glasses. *Journal of Non-Crystalline Solids*, 357(1), 88-95.

Cherif, A., Kanoun, A. & Maaref, H. 2011. Up-conversion Fluorescence Dynamics in Er^{3+}/Yb^{3+} Co-doped Tellurite Glasses. *Optica Applicata*, 41(1), 235-245. Retrieved from https://www.semanticscholar.org/paper/Up-conversion-fluorescence-dynamics-in-Er3-Yb3-co-Cherif-Kanoun/76df1e9baef857215a403b0ff95fdd3bce4797db on 20 April 2017.

Constantin, D. 2015. Why the Aspect Ratio? Shape Equivalence for the Extinction Spectra of Gold Nanoparticles. *The European Physical Journal E*, 38(11), 116.

de Araújo, C. B. & Kassab, L. R. P. 2017. Linear and Non-linear Optical Properties of Some Tellurium Oxide Glasses. *In* V.A.G. Rivera & D. Manzani (eds.). *Technological Advances in Tellurite Glasses: Properties, Processing, and Applications*, pp. 15-37. Cham, Switzerland: Springer.

Dellith, J. & Wendt, M. 2007. The M Emission Spectrum of ^{68}Erbium. *Microscopy and Microanalysis*, 13(3), 191-195.

Dellith, J. & Wendt, M. 2008. The M Emission Spectra of the Heavy Rare Earth Elements $67 \leq Z \leq 71$. *Microchimica Acta*, 161(3-4), 475-478.

UMS
UNIVERSITI MALAYSIA SABAH

Deslattes, R. D., Kessler, Jr., E. G., Indelicato, P., de Billy, L., Lindroth, E. & Anton, J. 2003. X-ray Transition Energies: New Approach to a Comprehensive Evaluation. *Reviews of Modern Physics*, 75(1), 35-99.

Devi, A. R. & Jayasankar, C.K. 1996. Optical Properties of Er^{3+} Ions in Lithium Borate Glasses and Comparative Energy Level Analyses of Er^{3+} Ions in Various Glasses. *Journal of Non-Crystalline Solids*, 197(2-3), 111-128.

Dieke, G. H. & Crosswhite, H. M. 1963. The Spectra of the Doubly and Triply Ionized Rare Earths. *Applied Optics*, 2(7), 675-686.

Dousti, M.R. 2013. Efficient Infrared-to-visible Upconversion Emission in Nd^{3+}-doped PbO-TeO_2 Glass Containing Silver Nanoparticles. *Journal of Applied Physics*, 114(11), 113105. Retrieved from https://www.researchgate.net/profile/M_Reza_Dousti/publication/257536342_Efficient_infrared-to-visible_upconversion_emission_in_Nd3-doped_PbO-TeO2_glass_containing_silver_NPs/links/553a45880cf245bdd762e1cb.pdf on 20 May 2018.

Dousti, M. R., Sahar, M.R., Amjad, R.J., Ghoshal, S.K. & Awang, A. 2013a. Surface Enhanced Raman Scattering and Up-conversion Emission by Silver Nanoparticles in Erbium-zinc-tellurite Glass. *Journal of Luminescence*, 143, 368-373.

Dousti, M. R., Sahar, M.R., Ghoshal, S.K., Amjad, R.J. & Samavati, A.R. 2013b. Effect of AgCl on Spectroscopic Properties of Erbium Doped Zinc Tellurite Glass. *Journal of Molecular Structure*, 1035, 6-12.

Dousti, M. R. & Hosseinian, S.R. 2014. Enhanced Upconversion Emission of Dy^{3+}-doped Tellurite Glass by Heat-treated Silver Nanoparticles. *Journal of Luminescence*, 154, 218-223.

Dousti, M. R., Amjad, R. J. & Mahraz, Z. A. S. 2015a. Enhanced Green and Red Upconversion Emissions in Er^{3+}-doped Boro-tellurite Glass Containing Gold Nanoparticles. *Journal of Molecular Structure*, 1079, 347-352.

Dousti, M.R., Amjad, R.J., Sahar, M.R., Zabidi, Z.M., Alias, A.N. & de Camargo, A.S.S. 2015b. Er^{3+}-doped Zinc Tellurite Glasses Revisited: Concentration Dependent Chemical Durability, Thermal Stability and Spectroscopic Properties. *Journal of Non-Crystalline Solids*, 429, 70-78.

Dousti, M.R. & Amjad, R.J. 2016a. Plasmon Assisted Luminescence in Rare Earth Doped Glasses. *In* C. D. Geddes (ed.). *Reviews in Plasmonics 2015*, pp. 340. Cham, Switzerland: Springer.

Dousti, M.R. & Amjad, R.J. 2016b. Plasmon Assisted Luminescence in Rare Earth Doped Glasses. *In* C. D. Geddes (ed.). *Reviews in Plasmonics 2015*, pp. 358. Cham, Switzerland: Springer.

UMS
UNIVERSITI MALAYSIA SABAH

Dousti, M.R. & Amjad, R.J. 2016c. Plasmon Assisted Luminescence in Rare Earth Doped Glasses. *In* C. D. Geddes (ed.). *Reviews in Plasmonics 2015*, pp. 339-345. Cham, Switzerland: Springer.

Dravid, V. 2016, December. *NUANCE Atomic and Nanoscale Characterization Experimental Center*. Poster session presented at the meeting of the National Science Foundation, Arlington, VA. Retrieved from http://www.nseresearch.org/2016/posters/NUANCE-Poster-2016.pdf on 20 June 2018.

Echlin, P. 2009. Sample Surface Charge Elimination. *In Handbook of Sample Preparation for Scanning Electron Microscopy and X-ray Microanalysis*, pp. 247-298. New York, NY: Springer.

Egerton, R.F. 2005a. The Transmission Electron Microscope. *In Physical Principles of Electron Microscopy: An Introduction to TEM, SEM, and AEM*, pp. 57-86. New York, NY: Springer.

Egerton, R.F. 2005b. The Scanning Electron Microscope. *In Physical Principles of Electron Microscopy: An Introduction to TEM, SEM, and AEM*, pp. 125-126. New York, NY: Springer.

Egerton, R.F. 2005c. The Scanning Electron Microscope. *In Physical Principles of Electron Microscopy: An Introduction to TEM, SEM, and AEM*, pp. 137. New York, NY: Springer.

Egerton, R. F. 2005d. Analytical Electron Microscopy. *In Physical Principles of Electron Microscopy: An Introduction to TEM, SEM, and AEM*, pp. 160. New York, NY: Springer.

El-Deen, L.M. S., Al Salhi, M.S. & Elkholy, M. M. 2008. IR and UV Spectral Studies for Rare Earth Doped Tellurite Glasses. *Journal of Alloys and Compounds*, 465(1-2), 333-339.

El-Mallawany, R. 2002a. Infrared and Raman Spectra of Tellurite Glasses. *In Tellurite Glass Handbook, Physical Properties and Data 1st Edition*, pp. 488-520. Boca Raton, Florida: CRC Press.

El-Mallawany, R. 2002b. Introduction to Tellurite Glasses. *In Tellurite Glasses: Physical Properties and Data 1st Edition*, pp. 17-39. Boca Raton, FL: CRC Press.

El-Mallawany, R. 2002c. Thermal Properties of Tellurite Glasses. *In Tellurite Glass Handbook, Physical Properties and Data 1st Edition*, pp. 204-205. Boca Raton, FL: CRC Press.

El-Mallawany, R. 2002d. Optical Properties of Tellurite Glasses in the Ultraviolet Region. *In Tellurite Glass Handbook, Physical Properties and Data 1st Edition*, pp. 436-442. Boca Raton, FL: CRC Press.

UMS
UNIVERSITI MALAYSIA SABAH

El-Mallawany, R. 2002e. Infrared and Raman spectra of Tellurite Glasses. *In Tellurite Glass Handbook, Physical Properties and Data 1st Edition*, pp. 481. Boca Raton, Florida: CRC Press.

El-Mallawany, R. 2002f. Infrared and Raman spectra of Tellurite Glasses. *In Tellurite Glass Handbook, Physical Properties and Data 1st Edition*, pp. 506-520. Boca Raton, Florida: CRC Press.

El-Mallawany, R., Dirar Abdalla, M. & Abbas Ahmed, I. 2008. New Tellurite Glass: Optical Properties. *Materials Chemistry and Physics*, 109(2-3), 291-296.

Ennos, A. E. 1953. The Origin of Specimen Contamination in the Electron Microscope. *British Journal of Applied Physics*, 4(4), 101-106.

Eraiah, B. 2017. Optical Properties of Silver-vanadium-phosphate Glasses. *Mapana Journal of Sciences*, 16(1), 1-7.

Erol, M., Küçükbayrak, S. & Ersoy-Meriçboyu, A. 2008. Comparison of the Properties of Glass, Glass-ceramic and Ceramic Materials Produced from Coal Fly Ash. *Journal of Hazardous Materials*, 153(1-2), 418-425.

Eustis, S. & El-Sayed, M. A. 2006. Why Gold Nanoparticles Are More Precious than Pretty Gold: Noble Metal Surface Plasmon Resonance and Its Enhancement of the Radiative and Nonradiative Properties of Nanocrystals of Different Shapes. *Chemical Society Reviews*, 35(3), 209-217.

Farahmandjou, M. & Salehizadeh, S. A. 2013. The Optical Band Gap and the Tailing States Determination in Glasses of TeO_2-V_2O_5-K_2O System. *Glass Physics and Chemistry*, 39(5), 473-479.

Fatimah, S., Sahar, M. R., Ghoshal, S. K., Ariffin, R. & Hamzah, K. 2014. Optical Absorption of Erbium Doped Tellurite Glass. *Advanced Materials Research*, 895, 245-249.

Faznny, M. F., Halimah, M. K. & Azlan, M. N. 2016. Effect of Lanthanum Oxide on Optical Properties of Zinc Borotellurite Glass System. *Journal of Optoelectronics and Biomedical Materials*, 8(2), 49-59.

Ferrell, S. K. 2015. *Effects of Metal and Semiconducting Nanoparticles on the Fluorescence and Optical Band Gap of Dy^{3+} Doped Lead Borate and Bismuth Borate Glasses*. (Master's Thesis) Western Illinois University. Retrieved from https://www.researchgate.net/publication/282735230_EFFECTS_OF_METAL_AND_SEMICONDUCTING_NPS_ON_THE_FLUORESCENCE_AND_OPTICAL_BAND_GAP_OF_Dy3_DOPED_LEAD_BORATE_AND_BISMUTH_BORATE_GLASSES?enrichId=rgreq-c29c92d6c100cbf29b037887116f7b61-XXX&enrichSource=Y292ZXJQYWdlOzI4MjczNTIzMDtBUzozMzAxODYxOTAwMTY1MTNAMTQ1NTczMzkzMDM1MQ%3D%3D&el=1_x_2&_esc=publicationCoverPdf on 15 July 2018.

Florez, A., Messaddeq, Y., Malta, O.L. & Aegerter, M.A. 1995. Optical Transition Probabilities and Compositional Dependence of Judd-Ofelt Parameters of

UMS
UNIVERSITI MALAYSIA SABAH

Er^{3+} Ions in Fluoroindate Glass. *Journal of Alloys and Compounds*, 227(2), 135-140.

Fultz, B. & Howe, J. M. 2013. Diffraction and X-ray Powder Diffractometer. *In Transmission Electron Microscopy and Diffractometry of Materials 4th Edition*, pp. 1-52. Berlin, Heidelberg: Springer.

Garcia, M.A. 2011. Surface Plasmons in Metal Nanoparticles: Fundamentals and Applications. *Journal of Physics D: Applied Physics*, 44(28), 283001.

Ghoshal, S.K., Awang, A., Sahar, M.R., Amjad, R.J. & Dousti, M.R. 2013. Spectroscopic and Structural Properties of TeO_2-ZnO-Na_2O-Er_2O_3-Au Glasses. *Chalcogenide Glasses*, 10(10), 411-420.

Ghoshal, S.K., Awang, A., Sahar, M.R. & Arifin, R. 2015. Gold Nanoparticles Assisted Surface Enhanced Raman Scattering and Luminescence of Er^{3+} Doped Zinc-sodium Tellurite Glass. *Journal of Luminescence*, 159, 265-273.

Goldstein, J., Newbury, D. E., Echlin, P., Joy, D. C., Romig. Jr., A. D., Lyman, C. E., Fiori, C. & Lifshin, E. 1992. Qualitative X-ray Analysis. *In Scanning Electron Microscopy and X-ray Microanalysis: A Text for Biologists, Materials Scientists, and Geologists 2nd Edition*, pp. 349-352. New York, NY: Plenum Press.

Goldstein, J. I., Newbury, D. E., Echlin, P., Joy, D. C., Lyman, C. E., Lifshin, E., Sawyer, L. & Michael, J. R. 2003a. Electron Beam-specimen Interactions. *In Scanning Electron Microscopy and X-ray Microanalysis 3rd Edition*, pp. 61-98. New York, NY: Kluwer Academic/Plenum Publishers.

Goldstein, J. I., Newbury, D. E., Echlin, P., Joy, D. C., Lyman, C. E., Lifshin, E., Sawyer, L. & Michael, J. R. 2003a. Electron Beam-specimen Interactions. *In Scanning Electron Microscopy and X-ray Microanalysis 3rd Edition*, pp. 75-98. New York, NY: Kluwer Academic/Plenum Publishers.

Goldstein, J. I., Newbury, D. E., Echlin, P., Joy, D. C., Lyman, C. E., Lifshin, E., Sawyer, L. & Michael, J. R. 2003b. Image Formation and Interpretation. *In Scanning Electron Microscopy and X-ray Microanalysis 3rd Edition*, pp. 99-103. New York, NY: Kluwer Academic/Plenum Publishers.

Goldstein, J. I., Newbury, D., Echlin, P., Joy, D. C., Lyman, C. E., Lifshin, E., Sawyer, L. & Michael, J. R. 2003c. Generation of X-rays in the SEM Specimen. *In Scanning Electron Microscopy and X-ray Microanalysis 3rd Edition*, pp. 274-281. New York, NY: Kluwer Academic/Plenum Publishers.

Goldstein, J. I., Newbury, D. E., Michael, J. R., Ritchie, N. W.M., Scott, J. H. J. & Joy, D. C. 2018a. Image Formation. *In Scanning Electron Microscopy and X-ray Microanalysis 4th Edition*, pp. 94. New York, NY: Springer.

UMS
UNIVERSITI MALAYSIA SABAH

Goldstein, J. I., Newbury, D. E., Michael, J. R., Ritchie, N. W.M., Scott, J. H. J. & Joy, D. C. 2018b. Energy Dispersive X-ray Spectrometry: Physical Principles and User-selected Parameters. *In Scanning Electron Microscopy and X-ray Microanalysis 4th Edition*, pp. 210-212. New York, NY: Springer.

Goldstein, J. I., Newbury, D. E., Michael, J. R., Ritchie, N. W.M., Scott, J. H. J. & Joy, D. C. 2018c. Image Defects. *In Scanning Electron Microscopy and X-ray Microscopy 4th Edition*, pp. 143. New York, NY: Springer.

Goldstein, J. I., Newbury, D. E., Michael, J. R., Ritchie, N. W.M., Scott, J. H. J. & Joy, D. C. 2018d. Energy Dispersive X-ray Spectrometry: Physical Principles and User-selected Parameters. *In Scanning Electron Microscopy and X-ray Microscopy 4th Edition*, pp. 213. New York, NY: Springer.

Goldstein, J. I., Newbury, D. E., Michael, J. R., Ritchie, N. W.M., Scott, J. H. J. & Joy, D. C. 2018e. Qualitative Elemental Analysis by Energy Dispersive X-ray Spectrometry. *In Scanning Electron Microscopy and X-ray Microscopy 4th Edition.*, pp. 266-272. New York, NY: Springer.

Goldstein, J. I., Newbury, D. E., Michael, J. R., Ritchie, N. W.M., Scott, J. H. J. & Joy, D. C. 2018f. Qualitative Elemental Analysis by Energy Dispersive X-ray Spectrometry. *In Scanning Electron Microscopy and X-ray Microscopy 4th Edition*, pp. 275-278. New York, NY: Springer.

Goldstein, J. I., Newbury, D. E., Michael, J. R., Ritchie, N. W.M., Scott, J. H. J. & Joy, D. C. 2018g. Qualitative Elemental Analysis by Energy Dispersive X-ray Spectrometry. *In Scanning Electron Microscopy and X-ray Microscopy 4th Edition*, pp. 281. New York, NY: Springer.

Goldstein, J. I., Newbury, D. E., Michael, J. R., Ritchie, N. W.M., Scott, J. H. J. & Joy, D. C. 2018h. X-rays. *In Scanning Electron Microscopy and X-ray Microscopy 4th Edition*, pp. 42-44. New York, NY: Springer.

Goodhew, P. J., Humphreys, J. & Beanland, R. 2001a. The Scanning Electron Microscope. *In Electron Microscopy and Analysis 3rd Edition*, pp. 122-131. London and New York: Taylor & Francis.

Goodhew, P. J., Humphreys, J. & Beanland, R. 2001b. Chemical Analysis in the Electron Microscope. *In Electron Microscopy and Analysis 3rd Edition*, pp. 169-173. London and New York: Taylor & Francis.

Goodhew, P. J., Humphreys, J. & Beanland, R. 2001c. Chemical Analysis in the Electron Microscope. *In Electron Microscopy and Analysis 3rd Edition*, pp. 180. London and New York: Taylor & Francis.

Gryczynski, Z., Matveeva, E. G., Calander, N., Zhang, J., Lakowicz, J. R., Gryczynski, I. 2007. Surface Plasmon Coupled Emission. *In* M. L.

UMS
UNIVERSITI MALAYSIA SABAH

Brongersma & P. G. Kik (eds.). *Surface Plasmon Nanophotonics*, pp. 247-248. Dordrecht, The Netherlands: Springer.

Halimah, M.K., Daud, W.M., Sidek, H.A.A., Zaidan, A.W. & Zainal, A.S. 2010. Optical Properties of Ternary Tellurite Glasses. *Materials Science-Poland*, 28(1), 173-180.

Han, M., Oh, S.J., Park, J. H. & Park, H. L. 1993. X-ray Photoelectron Spectroscopy Study of CaS:Eu and SrS:Eu Phosphors. *Journal of Applied Physics*. 73(9), 4546-4549.

Holton, I. 2012, May. Is Energy-dispersive Spectroscopy in the SEM a Substitute for Electron Probe Microanalysis? *Microscopy and Analysis, 26*(4): S4-S7 (AM). Retrieved from https://microscopy-analysis.com/magazine/issues/energy-dispersive-spectroscopy-sem-substitute-electron-probe-microanalysis on 20 June 2018.

Hu, Y., Qiu, J., Song, Z. & Zhou, D. 2014. Ag_2O Dependent Up-conversion Luminescence Properties in $Tm^{3+}/Er^{3+}/Yb^{3+}$ Co-doped Oxyfluorogermanate Glasses. *Journal of Applied Physics*, 115(8), 083512-1 - 6.

Huang, Y.D., Mortier, M. & Auzel, F. 2001a. Stark Levels Analysis for Er^{3+}-doped Oxide Glasses: Germanate and Silicate. *Optical Materials*, 15(4), 243-260.

Huang, Y.D., Mortier, M. & Auzel, F. 2001b. Stark Level Analysis for Er^{3+}-doped ZBLAN Glass. *Optical Materials*, 17(4), 501-511.

Jaba, N., Mermet, A., Duval, E. & Champagnon, B. 2005. Raman Spectroscopy of Er^{3+}-doped Zinc Tellurite Glasses. *Journal of Non-Crystalline Solids*, 351(10-11), 833-837.

Jain, P. K., Huang, X., El-Sayed, I. H. & El-Sayed, M. A. 2007. Review of Some Interesting Surface Plasmon Resonance-enhanced Properties of Noble Metal Nanoparticles and Their Applications to Biosystems. *Plasmonics*, 2(3), 107-118.

Jha, A., Shen, S. & Naftaly, M. 2000. Structural Origin of Spectral Broadening of 1.5-μm Emission in Er^{3+}-doped Tellurite Glasses. *Physical Review B*, 62(10), 6215-6227.

Jlassi, I., Elhouichet, H. & Ferid, M. 2011. Thermal and Optical Properties of Tellurite Glasses Doped Erbium. *Journal of Materials Science*, 46(3), 806-812.

Jørgensen, C. K. & Judd, B. R. 1964. Hypersensitive Pseudoquadrupole Transitions in Lanthanides. *Molecular Physics: An International Journal at the Interface between Chemistry and Physics*, 8(3), 281-290.

Jose, R. & Ohishi, Y. 2007. Ultra-broadband Raman Gain Media for Photonics Device Applications. *Optical Components and Materials IV*, 6469, 64690K.

Kamitsos, E.I. 2015. Infrared Spectroscopy of Glasses. *In* M. Affatigato (ed.). *Modern Glass Characterization*, pp. 77-78. Hoboken, NJ: John Wiley & Sons.

UMS
UNIVERSITI MALAYSIA SABAH

Karmakar, B., Som, T., Singh, S. P. & Nath, M. 2010. Nanometal-glass Hybrid Nanocomposites: Synthesis, Properties and Applications. *Transactions of the Indian Ceramic Society*, 69(3), 171-186.

Karmakar, B. 2016a. Fundamentals of Glass and Glass Nanocomposites. *In* B. Karmakar, K. Rademann & A. L. Stepanov (eds.). *Glass Nanocomposites: Synthesis, Properties and Applications 1st Edition*, pp. 6-12. Oxford, UK & Cambridge, USA: Elsevier.

Karmakar, B. 2016b. Fundamentals of Glass and Glass Nanocomposites. *In* B. Karmakar, K. Rademann & A. L. Stepanov (eds.). *Glass Nanocomposites: Synthesis, Properties and Applications 1st Edition*, pp. 21-22. Oxford, UK & Cambridge, USA: Elsevier.

Karmakar, B. 2016c. Fundamentals of Glass and Glass Nanocomposites. *In* B. Karmakar, K. Rademann & A. L. Stepanov (eds.). *Glass Nanocomposites: Synthesis, Properties and Applications 1st Edition*, pp. 29-34. Oxford, UK & Cambridge, USA: Elsevier.

Karmakar, B. 2016d. Fundamentals of Glass and Glass Nanocomposites. *In* B. Karmakar, K. Rademann & A. L. Stepanov (eds.). *Glass Nanocomposites: Synthesis, Properties and Applications 1st Edition*, pp. 16-17. Oxford, UK & Cambridge, USA: Elsevier.

Karmakar, B. 2016e. Fundamentals of Glass and Glass Nanocomposites. *In* B. Karmakar, K. Rademann & A. L. Stepanov (eds.). *Glass Nanocomposites: Synthesis, Properties and Applications 1st Edition*, pp. 38-41. Oxford, UK & Cambridge, USA: Elsevier.

Kenyon, A.J. 2002. Recent Developments in Rare-earth Doped Materials for Optoelectronics. *Progress in Quantum Electronics*, 26(4-5), 225-284.

Khondker, A. & Lakhani, S. 2015. X-Ray Diffraction: A Comprehensive Explanation for Multipurpose Research. *International Journal of Interdisciplinary Research and Innovations*, 3(1), 60-64. Retrieved from http://www.researchpublish.com/journal/IJIRI/Issue-1-January-2015-March-2015/0 on 20 May 2018.

Klinger, M.I. 2013. General Description of Glasses and Glass Transition. *In Glassy Disordered Systems: Glass Formation and Universal Anomalous Low-Energy Properties*, pp. 5-6. Toh Tuck Link, Singapore: World Scientific.

Kortright, J. B. & Thompson, A. C. 2009. X-ray Properties of Elements: X-ray Emission Energies. *In* A. C. Thompson (ed.), *X-ray Data Booklet 3rd Edition*, 1-8 – 1-27.

Krishnaiah, K. V., Marques-Hueso, J. & Kashyap, R. 2017. Broadband Emission in Tellurite Glasses. *In* V.A.G. Rivera & D. Manzani (eds.). *Technological*

UMS
UNIVERSITI MALAYSIA SABAH

Advances in Tellurite Glasses: Properties, Processing, and Applications, pp. 158-163. Cham, Switzerland: Springer.

Krishnaiah, K. V., Venkataiah, G., Marques-Hueso, J., Dharmaiah, P., Jayasankar, C.K. & Richards, B.S. 2015. Broadband Near-infrared Luminescence and Visible Upconversion of Er^{3+}-doped Tungstate-tellurite Glasses. *Science of Advanced Materials*, 7(2), 345-353.

Kumar, K., Rai, S.B. & Rai, D.K. 2006. Upconversion Studies in Er^{3+} Doped TeO_2-M_2O (M=Li, Na and K) Binary Glasses. *Solid State Communications*, 139(7), 363-369.

Kumar, S. & Sood, A.K. 2016. Ultrafast Response of Plasmonic Nanostructures. *In* C. D. Geddes (ed.). *Reviews in Plasmonics 2015*, pp. 136. Cham, Switzerland: Springer.

Lakowicz, J. R. 2006a. Instrumentation for Fluorescence Spectroscopy. *In Principles of Fluorescence Spectroscopy 3rd Edition*, pp. 27-28. New York, NY: Springer.

Lakowicz, J. R. 2006b. Instrumentation for Fluorescence Spectroscopy. *In Principles of Fluorescence Spectroscopy 3rd Edition*, pp. 30-31. New York, NY: Springer.

Lakshminarayana, G., Kaky, K.M., Baki, S.O., Ye, S., Lira, A., Kityk, I.V. & Mahdi, M.A. 2016. Concentration Dependent Structural, Thermal, and Optical Features of Pr^{3+}-doped Multicomponent Tellurite Glasses. *Journal of Alloys and Compounds*, 686, 769-784.

Lakshminarayana, G., Kaky, K.M., Baki, S.O., Lira, A., Nayar, P., Kityk, I.V. & Mahdi, M.A. 2017. Physical, Structural, Thermal, and Optical Spectroscopy Studies of TeO_2-B_2O_3-MoO_3-ZnO-R_2O (R=Li, Na, and K)/MO (M = Mg, Ca, and Pb) Glasses. *Journal of Alloys and Compounds*, 690, 799-816.

Lee, D. K., Park, S. I., Lee, J. K. & Hwang, N. M. 2007. A Theoretical Model for Digestive Ripening. *Acta Materialia*, 55(15), 5281-5288.

Lee, D. K. & Hwang, N. M. 2009. Thermondynamics and Kinetics of Monodisperse Alloy Nanoparticles Synthesized through Digestive Ripening. *Scripta Materialia*, 61(3), 304-307.

Lin, X. M., Sorensen, C. M. & Klabunde, K. J. 2000. Digestive Ripening, Nanophase Segregation and Superlattice Formation in Gold Nanocrystal Colloids. *Journal of Nanoparticle Research*, 2(2), 157-164.

Lin, H., Zhang, Y.Y. & Pun, E.Y. 2008. Fluorescence Investigation of Ho^{3+} in Yb^{3+} Sensitized Mixed-alkali Bismuth Gallate Glass. *Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy*, 71(4), 1547-1550.

UMS
UNIVERSITI MALAYSIA SABAH

Long, D.A. 2002a. Survey of Light-scattering Phenomena. *In The Raman Effect: A Unified Treatment of the Theory of Raman Scattering by Molecules*, pp. 3-18. West Sussex, England: John Wiley and Sons.

Long, D. A. 2002b. Classical Theory of Rayleigh and Raman Scattering. *In The Raman Effect: A Unified Treatment of the Theory of Raman Scattering by Molecules*, pp. 47-48. West Sussex, England: John Wiley and Sons.

Lövestam, G., Rauscher, H., Roebben, G., Klüttgen, B. S., Gibson, N., Putaud, J. & Stamm, H. 2010. *Considerations on a Definition of Nanomaterial for Regulatory Purposes*. JRC Publication No. JRC58726. European Union: Joint Research Centre. Retrieved from https://ec.europa.eu/jrc/en/publication/reference-reports/considerations-definition-nanomaterial-regulatory-purposes on 20 August 2018.

Maheshvaran, K., Veeran, P.K. & Marimuthu, K. 2013. Structural and Optical Studies on Eu^{3+} Doped Boro Tellurite Glasses. *Solid State Sciences*, 17, 54-62.

Malta, O.L., Santa-Cruz, P.A., De Sa, G.F. & Auzel, F. 1985. Fluorescence Enhancement Induced by the Presence of Small Silver Particles in Eu^{3+} Doped Materials. *Journal of Luminescence*, 33(3), 261-272.

Malta, O. L. & Carlos, L. D. 2003. Intensities of 4f-4f Transitions in Glass Materials. *Química Nova*, 26(6), 889-895.

Marjanovic, S., Toulouse, J., Jain, H., Sandmann, C., Dierolf, V., Kortan, A.R., Kopylov, N. & Ahrens, R.G. 2003. Characterization of New Erbium-doped Tellurite Glasses and Fibers. *Journal of Non-Crystalline Solids*, 322(1-3), 311-318.

McClure, D. S. & Kiss, Z. 1963. Survey of the Spectra of the Divalent Rare-earth Ions in Cubic Crystals. *Journal of Chemical Physics*, 39(12), 3251-3257.

Mitra, A. & De, G. 2016. Sol-gel Synthesis of Metal Nanoparticle Incorporated Oxide Films on Glass. *In* B. Karmakar, K. Rademann & A. L. Stepanov (eds.). *Glass – Nanocomposites: Synthesis, Properties and Applications 1st Edition*, pp. 147. Oxford, UK & Cambridge, USA: Elsevier.

Munoz-Martín, D., Villegas, M.A., Gonzalo, J. & Fernández-Navarro, J.M. 2009. Characterisation of Glasses in the TeO_2-WO_3-PbO System. *Journal of the European Ceramic Society*, 29(14), 2903-2913.

Murugan, G. S., Suzuki, T. & Ohishi, Y. 2006. Raman Characterisitics and Nonlinear Optical Properties of Tellurite and Phosphotellurite Glasses Containing Heavy Metal Oxides with Ultrabroad Raman Bands. *Journal of Applied Physics Letters*, 100(2), 023107.

Mustafa, I. S., Razali, N. A. N., Azman, N. Z. N., Yahaya, N. Z., Zaini, M. Z. M., Rusli, N. L., Nizamani, M. B. & Halimah, M. K. 2017. Comprehensive Study

UMS
UNIVERSITI MALAYSIA SABAH

of Electronic Polarizability and Band Gap of B_2O_3–Bi_2O_3–ZnO–SiO_2 Glass Network. *Journal of Advanced Dielectrics*, 7(5), 1750031.

Nandi, P. & Jose, G. 2006. Spectroscopic Properties of Er^{3+} Doped Phospho-tellurite Glasses. *Physica B: Condensed Matter*, 381(1-2), 66-72.

Nazabal, V., Todoroki, S., Nukui, A., Matsumoto, T., Suehara, S., Hondo, T., Araki, T., Inoue, S., Rivero, C. & Cardinal, T. 2003. Oxyfluoride Tellurite Glasses Doped by Erbium: Thermal Analysis, Structural Organization and Spectral Properties. *Journal of Non-Crystalline Solids*, 325(1-3), 85-102.

Newbury, D. E. & Ritchie, N. W. M. 2013. Is Scanning Electron Microscopy/Energy-Dispersive X-ray Spectrometry (SEM/EDS) Quantitative? *Scanning*, 35(3), 141-168.

Newbury, D. E. & Ritchie, N. W. M. 2015. Performing Elemental Microanalysis with High Accuracy and High Precision by Scanning Electron Microscopy/Silicon Drift Detector Energy-dispersive X-ray Spectrometry (SEM-SDD-EDS). *Journal of Materials Science*, 50(2), 493-518.

Nurhafizah, H., Rohani, M.S. & Ghoshal, S.K. 2017.Self Cleanliness of Er^{3+}/Nd^{3+} Co-doped Lithium Niobate Tellurite Glass Containing Silver Nanoparticles. *Journal of Non-Crystalline Solids*, 455, 62-69.

O'Donnell, M. D., Miller, C. A., Furniss, D., Tikhomirov, V. K. & Seddon, A. B. 2003. Fluorotellurite Glasses with Improved Mid-infrared Transmission. *Journal of Non-Crystalline Solids*, 331(1-3), 48-57.

Oo, H. M., Mohamed-Kamari, H., Mohd Daud Wan-Yusoff, W. 2012. Optical Properties of Bismuth Tellurite Based Glass. *International Journal of Molecular Sciences*, 13(4), 4623-4631.

Oxford Instruments Analytical Limited. 2008. *X-Max: The Largest SDD - Size It Matters*. High Wycombe, UK: Oxford Instruments. Retrieved from https://aiim.uow.edu.au/content/groups/public/@web/@aiim/documents/doc/uow154676.pdf on 10 June 2018.

Oxford Instruments Analytical Limited. 2012. *Silicon Drift Detectors Explained*. High Wycombe, UK: Oxford Instruments. Retrieved from https://nano.oxinst.com/products/x-max/x-max on 10 June 2018.

Pan, Z., Ueda, A., Aga Jr., R. Burger, A., Mu, R. & Morgan, S. H. 2010. Spectroscopic Studies of Er^{3+} Doped Ge-Ga-S Glass Containing Silver Nanoparticles. *Journal of Non-Crystalline Solids*, 356(23-24), 1097-1101.

Parveen, N., Jali, V. M. & Patil, S. D. 2016. Structure and Optical Properties of TeO_2-GeO_2 Glasses. *International Journal of Advances in Science Engineering and Technology*, 4(1), 89-94. Retrieved from http://iraj.in/journal/IJASEAT/paper_detail.php?paper_id=3867&name=Structure_And_Optical_Properties_Of_Teo2-Geo2_Glasses on 20 May 2018.

UMS
UNIVERSITI MALAYSIA SABAH

Pavani, P.G., Sadhana, K. & Mouli, V.C. 2011. Optical, Physical and Structural Studies of Boro-zinc Tellurite Glasses. *Physica B*, 406(6-7), 1242-1247.

Perkins, S. T., Cullen, D. E., Chen, M. H., Hubbell, J. H., Rathkopf, J. & Scofield, J. 1991. *Tables and Graphs of Atomic Subshell and Relaxation Data Derived from the LLNL Evaluated Atomic Data Library (EADL), Z = 1 – 100.* Report No. UCRL-50400-Vol. 30. Oak Ridge, USA: Office of Scientific and Technical Information. Retrieved from https://www.osti.gov/biblio/10121422-tables-graphs-atomic-subshell-relaxation-data-derived-from-llnl-evaluated-atomic-data-library-eadl on 20 June 2018.

Pellerin, N., Blondeau, J-P., Noui, S., Allix, M., Ory, S., Veron, O., De Sousa Meneses, D. & Massiot. Control of Selective Silicate Glass Coloration by Gold Metallic Nanoparticles: Structural Investigation, Growth Mechanisms, and Plasmon Resonance Modelization. *Gold Bulletin*, 46(4), 243-255.

Poppe, L.J., Paskevich, V.F., Hathaway, J.C. & Blackwood, D.S. 2001. *A Laboratory Manual for X-Ray Powder Diffraction.* Report 01-041. Woods Hole, MA: U.S. Geological Survey. Retrieved from https://www.researchgate.net/publication/265108796_A_Laboratory_Manual_for_X-Ray_Powder_Diffraction on 20 May 2018.

Postek, M. T., Vladár, A. E. & Purushotham, K. P. 2014. Does Your SEM Really Tell the Truth? How Would You Know? Part 2. *Scanning*, 36(3), 347-355.

Pyrz, W. D. & Buttrey, D. J. 2008. Particle Size Determination using TEM: A Discussion of Image Acquisition and Analysis for the Novice Microscopist. *Langmuir*, 24(20), 11350-11360.

Quorum Technologies Limited Company. no yr. *Q150T Turbomolecular-pumped Coating System: High Vacuum Sputtering, Carbon and Metal Evaporation for SEM, TEM and Thin Film Applications.* East Sussex, England: Quorum Technologies. Retrieved from https://www.quorumtech.com/__assets__/pdf/product/Q150T-brochure-final.pdf on 20 May 2018.

Ramer, G. & Lendl, B. 2013. Attenuated Total Reflection Fourier Transform Infrared Spectroscopy. *In Encyclopedia of Analytical Chemistry: Applications, Theory and Instrumentation,* pp. 1-27. Retrieved from https://onlinelibrary.wiley.com/doi/abs/10.1002/9780470027318.a9287 on 20 May 2018.

Reisfeld, R., Katz, G., Spector, N., Jørgensen, C. K., Jacobani, C. & de Pape, R. 1982. Optical Transition Probabilities of Er^{3+} in Fluoride Glasses. *Journal of Solid State Chemistry*, 41(3), 253-261.

Rivera, V. A. G., Osorio, S. P. A., Ledemi, Y., Manzani, D., Messadeq, Y., Nunes, L.A.O. & Marega Jr., E. 2010. Localized Surface Plasmon Resonance

UMS
UNIVERSITI MALAYSIA SABAH

Interaction with Er^{3+}-doped Tellurite Glass. *Optics Express*, 18(24), 25321-25328.

Rivera, V. A. G., Manzani, D., Messaddeq, Y., Nunes, L. A. O. & Marega Jr., E. 2011. Study of Er^{3+} Fluorescence on Tellurite Glasses Containing Ag Nanoparticles. *Journal of Physics: Conference Series*, 274(1), 012123.

Rivera, V.A.G., Ferri, F.A. & Marega Jr, E. 2012. Localized Surface Plasmon Resonances: Noble Metal Nanoparticle Interaction with Rare-earth Ions. *In* K. Y. Kim (ed.). *Plasmonics - Principles and Applications*, pp. 283-312. Rijeka, Croatia: InTech. doi: 10.5772/50753. Retrieved from https://www.intechopen.com/books/plasmonics-principles-and-applications/localized-surface-plasmon-resonances-noble-metal-nanoparticle-interaction-with-rare-earth-ions on 20 May 2018.

Rivera, V. A. G., Ledemi, Y., Osorio, S. P. A., Manzani, D., Ferri, F. A., Ribeiro, S. J.L., Nunes, L. A. O. & Marega Jr., E. 2013. Tunable Plasmon Resonance Modes on Gold Nanoparticles in Er^{3+}-doped Germanium-tellurite Glass. *Journal of Non-Crystalline Solids*, 378, 126-134.

Rivera, V. A. G., Ledemi, Y., Pereira-da-Silva, M. A., Messaddeq, Y., & Marega Jr, E. 2016. Plasmon-photon Conversion to Near-infrared Emission from Yb^{3+}: (Au/Ag-nanoparticles) in Tungsten-tellurite Glasses. *Scientific Reports*, 6, 18464.

Roy, B., Jain, H., Saha, S.K. & Chakravorty, D. 1995. Comparison of Structure of Alkali Silicate Glasses Prepared by Sol-gel and Melt-quench Methods. *Journal of Non-Crystalline Solids*, 183(3), 268-276.

Ru, E. L. C. & Etchegoin, P. G. 2009a. A Quick Overview of Surface-enhanced Raman Spectroscopy. *In Principles of Surface-enhanced Raman Spectroscopy and Related Plasmonic Effects 1st Edition*, pp. 1-3. Amsterdam, The Netherlands: Elsevier. Retrieved from http://www.fulviofrisone.com/attachments/article/406/principles%20of%20surface%20enhanced%20raman%20spectroscopy0444527796.pdf on 20 April 2018.

Ru, E. L. C. & Etchegoin, P. G. 2009b. EM Enhancements and Plasmon Resonances: Examples and Discussion. *In Principles of Surface-enhanced Raman Spectroscopy and Related Plasmonic Effects 1st Edition*, pp. 348. Amsterdam, The Netherlands: Elsevier. Retrieved from http://www.fulviofrisone.com/attachments/article/406/principles%20of%20surface%20enhanced%20raman%20spectroscopy0444527796.pdf on 20 April 2018.

Sahar, M. R. & Ghoshal, S. K. 2015. Nanoglass: Present Challenges and Future Promises. *Advanced Materials Research*, 1108, 45-58. doi:10.4028/www.scientific.net/AMR.1108.45

UMS
UNIVERSITI MALAYSIA SABAH

Sakka, S. 2014. Sol-gel Glasses and Their Future Applications. *Transaction of the Indian Ceramic Society*, 46(1), 1-11. doi: 10.1080/0371750X.1987.10822820

Sangwarantee, N., Kaewkhao, J., & Chanthima, N. 2016. Visible Luminescence Properties of Sm^{3+} Doped Magnesium Bismuth Phosphate Glasses. *Key Engineering Materials*, 675-676, 405-408.

Selvaraju, K. & Marimuthu, K. 2012. Structural and Spectroscopic Studies on Concentration Dependent Er^{3+} Doped Boro-tellurite Glasses. *Journal of Luminescence*, 132(5), 1171-1178.

Selvaraju, K., Vijaya, N., Marimuthu, K. & Lavin, V. 2013. Composition Dependent Spectroscopic Properties of Er^{3+}-doped Boro-tellurite Glasses. *Physica Status Solidi A*, 210(3), 607-615.

Shirley, G.Z. 2014, August, 26. Need help on lognormal function fitting [Online forum post]. Message posted to https://www.originlab.com/forum/topic.asp?TOPIC_ID=19913

Shelby, J.E. 2005. Introduction. *In Introduction to Glass Science and Technology 2nd Edition*, pp. 1-3. Cambridge, UK: The Royal Society of Chemistry.

Shelby, J.E. 2005. Glass Melting. *In Introduction to Glass Science and Technology 2nd Edition*, pp. 26-48. Cambridge, UK: The Royal Society of Chemistry.

Sidek, H.A.A., Rosmawati, S., Talib, Z.A., Halimah, M.K. & Daud, W.M. 2009. Synthesis and Optical Properties of ZnO-TeO_2 Glass System. *American Journal of Applied Sciences*, 6(8), 1489-1494.

Smith, E. and Dent, G. 2005. Introduction, Basic Theory and Principles. *In Modern Raman Spectroscopy – A Practical Approach*, pp. 1-4. West Sussex, England: John Wiley & Sons.

Smith, B. C. 2011a. Introduction to Infrared Spectroscopy. *In Fundamentals of Fourier Transform Infrared Spectroscopy 2nd Edition*, pp. 5-6. Boca Raton, FL: CRC Press.

Smith, B. C. 2011b. How an FTIR Works. *In Fundamentals of Fourier Transform Infrared Spectroscopy 2nd Edition*, pp. 19-27. Boca Raton, FL: CRC Press.

Smith, B. C. 2011c. Preparing Samples Properly. *In Fundamentals of Fourier Transform Infrared Spectroscopy 2nd Edition*, pp. 129-144. Boca Raton, FL: CRC Press.

Solé, J. G., Bausa, L.E. & Jaque, D. 2005a. Fundamentals. *In An Introduction to Optical Spectroscopy of Inorganic Solids*, pp. 8-15. West Sussex, England: John Wiley & Sons.

UMS
UNIVERSITI MALAYSIA SABAH

Solé, J. G., Bausá, L.E. & Jaque, D. 2005b. Fundamentals. *In An Introduction to Optical Spectroscopy of Inorganic Solids*, pp. 16-18. West Sussex, England: John Wiley & Sons.

Solé, J. G., Bausá, L.E. & Jaque, D. 2005c. Applications: Rare Earth and Transition Metal Ions, and Color Centers. *In An Introduction to Optical Spectroscopy of Inorganic Solids*, pp. 200-201. West Sussex, England: John Wiley & Sons.

Soltani, I., Hraiech, S., Horchani-Naifer, K., Elhouichet, H., Gelloz, B. & Férid, M. 2016. Growth of Silver Nanoparticles Stimulate Spectroscopic Properties of Er^{3+}-doped Phosphate Glasses: Heat Treatment Effect. *Journal of Alloys and Compounds*, 686, 556-563.

Som, T. & Karmakar, B. 2008. Infrared-to-red Upconversion Luminescence in Samarium-doped Antimony Glasses. *Journal of Luminescence*, 128(12), 1989-1996.

Som, T. & Karmakar, B. 2009a. Nanosilver Enhanced Upconversion Fluorescence of Erbium Ions in Er^{3+}: Ag-antimony Glass Nanocomposites. *Journal of Applied Physics*, 105(1), 013102.

Som, T. & Karmakar, B. 2009b. Enhancement of Er^{3+} Upconverted Luminescence in Er^{3+}: Au-antimony Glass Dichroic Nanocomposites Containing Hexagonal Au Nanoparticles. *Journal of the Optical Society of America B*, 26(12), B21-B27.

Som, T. & Karmakar, B. 2009c. Core-shell Au-Ag Nanoparticles in Dielectric Nanocomposites with Plasmon-enhanced Fluorescence: A New Paradigm in Antimony Glasses. *Nano Research*, 2(8), 607-616.

Som, T. & Karmakar, B. 2009d. Efficient Green and Red Fluorescence Upconversion in Erbium Doped New Low Phonon Antimony Glasses. *Optical Materials*, 31(4), 609-618.

Som, T. & Karmakar, B. 2010a. Surface Plasmon Resonance and Enhanced Fluorescence Application of Single-step Synthesized Elliptical Nano Gold-embedded Antimony Glass Dichroic Nanocomposites. *Plasmonics*, 5(2), 149-159.

Som, T. & Karmakar, B. 2010b. Enhanced Frequency Upconversion of Sm^{3+} Ions by Elliptical Au Nanoparticles in Dichroic Sm^{3+}: Au-antimony Glass Nanocomposites. *Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy*, 75(2), 640-646.

Som, T. & Karmakar, B. 2011. Nano Silver: Antimony Glass Hybrid Nanocomposites and Their Enhanced Fluorescence Application. *Solid State Sciences*, 13(5), 887-895.

Som, T., Singh, S. P. & Karmakar, B. 2016. Plasmonic Antimony and Bismuth Oxide Glass: Synthesis and Enhanced Photoluminescence. *In* B. Karmakar, K. Rademann & A. L. Stepanov (eds.). *Glass Nanocomposites: Synthesis, Properties and Applications 1st Edition*, pp.225. Cambridge, MA: Elsevier.

UMS
UNIVERSITI MALAYSIA SABAH

Statham, P. J. 2002. Limitations to Accuracy in Extracting Characteristic Line Intensities from X-ray Spectra. *Journal of Research of the National Institute of Standards and Technology*, 107(6), 531-546.

Studenyak, I., Kranjčec, M. & Kurik, M. 2014. Urbach Rule in Solid State Physics. *International Journal of Optics and Applications*, 4(3), 76-83.

Sur, U. K. 2017. Surface-enhanced Raman scattering, Raman Spectroscopy and Applications. *In* K. Maaz (ed.). *Raman Spectroscopy and Applications*, pp. 293-295. Rijeka, Croatia: InTech. doi: 10.5772/66084. Retrieved from https://www.intechopen.com/books/raman-spectroscopy-and-applications/surface-enhanced-raman-scattering on 10 April 2019.

Suryanarayana, C. & Norton, M. G. 1998a. Lattices and Crystal Structures. *In X-ray Diffraction: A Practical Approach 1st Edition*, pp. 60-61. New York, NY: Springer.

Suryanarayana, C. & Norton, M. G. 1998b. Lattices and Crystal Structures. *In X-ray Diffraction: A Practical Approach 1st Edition*, pp. 50-62. New York, NY: Springer.

Tarafder, A. 2015. *Processing and Characterization of Au^0 and Au^0-Er^{3+} Containing Plasmonic Glass Nanocomposites in K_2O-ZnO-SiO_2 Glass System*. (PhD thesis). Retrieved from http://shodhganga.inflibnet.ac.in/bitstream/10603/163894/17/17_chapter%208.pdf on 20 June 2018.

Thanh, N. T. K., Maclean, N. & Mahiddine, S. 2014. Mechanisms of Nucleation and Growth of Nanoparticles in Solution. *Chemical Reviews*, 114(15), 7610-7630.

Triola, M. F. 2010. Statistics for Describing, Exploring and Comparing Data. *In Elementary Statistics Technology Update 11th Edition*, pp. 92. Boston, MA: Pearson.

Van Deun, R., Binnemans, K., Görller-Warland, C. & Adam, J.L. 1999. Spectroscopic Properties of Trivalent Samarium Ions in Glasses. *Proceedings Volume 3622, Rare-Earth-Doped Materials and Devices III*. January, 1999. San Jose, California. Retrieved from https://www.researchgate.net/profile/Rik_Deun/publication/230687749_Spectroscopic_properties_of_trivalent_samarium_ions_in_glasses/links/0c9605168445260b99000000/Spectroscopic-properties-of-trivalent-samarium-ions-in-glasses.pdf on 26 June 2019.

Vijayakumar, R. & Marimuthu, K. 2016. Luminescence Studies on Ag Nanoparticles Embedded Eu^{3+} Doped Boro-phosphate Glasses. *Journal of Alloys and Compounds*, 665, 294-303.

UMS
UNIVERSITI MALAYSIA SABAH

Voorhess, P. W. 1985. The Theory of Ostwald Ripening. *Journal of Statistical Physics*, 38(1-2), 231-252.

Wang, Z. L. no yr. *Fundamental Theory of Transmission Electron Microscope*. Retrieved from http://www.nanoscience.gatech.edu/zlwang/research/tem.html on 10 April 2018.

Wang, J.S., Vogel, E.M. & Snitzer, E. 1994. Tellurite Glass: A New Candidate for Fiber Devices. *Optical Materials*, 3(3), 187-203.

Wang, Y., Deng, J., Di, J. & Tu, Y. 2009. Electrodeposition of Large Size Gold Nanoparticles on Indium Tin Oxide Glass and Application as Refractive Index Sensor. *Electrochemistry Communications*, 11(5), 1034-1037.

Wang, C., Singh, P., Kim, Y. J., Mathiyalagan, R., Myagmarjav, D., Wang, D., Jin, C. & Yang, D. C. 2015. Characterization and Antimicrobial Application of Biosynthsized Gold and Silver Nanoparticles by using *Microbacterium resistens*. *Artificial Cells, Nanomedicine, and Biotechnology*, 44(7), 1714-1721.

Warren, B. E. 1940. X-ray Diffraction Study of the Structure of Glass. *Chemical Reviews*, 26(2), 237-255.

Wassilkowska, A., Czaplicka-Kotas, A., Bielski, A. & Zielina, M. 2014. An Analysis of the Elemental Composition of Micro-samples using EDS Technique. *Czasopismo Techniczne*, Chemia Zeszyt 1-Ch (18) 2014, 133-148.

Williams, D. B. & Carter, C. B. 2009a. The Transmission Electron Microscope. *In Transmission Electron Microscopy: A Textbook for Materials Science*, pp. 3-4. New York, NY: Springer.

Williams, D. B. & Carter, C. B. 2009b. The Transmission Electron Microscope. *In Transmission Electron Microscopy: A Textbook for Materials Science*, pp. 5-14. New York, NY: Springer.

Williams, D. B. & Carter, C. B. 2009c. Scattering and Diffraction. *In Transmission Electron Microscopy: A Textbook for Materials Science*, pp. 24-26. New York, NY: Springer.

Xu, H. & Jiang, Z. 2003. Dynamics of Visible-to-ultraviolet Upconversion in $YAlO_3$: 1% Er^{3+}. *Chemical Physics*, 287(1-2), 155-159.

Yang, J., Zhang, L., Wen, L. Dai, S., Hu, L. & Jiang, Z. 2004. Optical Transitions and Upconversion Luminescence of Er^{3+}/Yb^{3+}-codoped Halide Modified Tellurite Glasses. *Journal of Applied Physics*, 95(6), 3020-3026.

UMS
UNIVERSITI MALAYSIA SABAH

Yang, J., Yang, Z., Wang, Y., Qiu, J. & Song, Z. 2016. Upconversion Luminescence Enhancement of SiO_2:Yb^{3+}, Tb^{3+} Inverse Opal Photonic Crystal by Gold Nanoparticles. *Journal of Non-Crystalline Solids*, 437, 53-57.

Yao, J. H., Elder, K. R., Guo, H. & Grant, M. 1993. Theory and Simulation of Ostwald Ripening. *Physical Review B*, 47(21), 14110-14125.

Yen, W. M. 1986. Optical Spectroscopy of Ions in Inorganic Glasses. *In* I. Zschokke (ed.). *Optical Spectroscopy of Inorganic Glasses*, pp. 40-43. Dordrecht, Holland: D. Reidel Publishing Company.

Yoko, T., Kamiya, K., Yamada, H. & Tanaka, K. 1998. Glass-forming Region and Structure of Oxyhalide Glasses in the system LiCl-Li_2O-TeO_2. *Communications of the American Ceramic Society*, 71(2), C-70 – C-71.

Yousef, E. L. S, Hegazy, H. H., Elokr, M. M. & Aboudeif, Y. M. 2015. Raman Spectroscopy and Raman Gain Coefficient of Telluroniobium-zinc-lead Oxyglasses Doped with Rare Earth. *Chalcogenide Letters*, 12(12), 653-663. Retrieved from http://www.chalcogen.ro/653_ElSayed.pdf on 10 May 2018.

Yusof, N. N., Ghoshal, S. K., Ariffin, R. & Sahar, M. R. 2015. Modified Absorption Features of Titania-erbium Incorporated Plasmonic Tellurite Glass System. *Jurnal Teknologi*, 76(13), 89-94.

Yusof, N.N., Ghoshal, S.K. & Azlan, M.N. 2017. Optical Properties of Titania Nanoparticles Embedded Er^{3+}-doped Tellurite Glass: Judd-Ofelt Analysis. *Journal of Alloys and Compounds*, 724, 1083-1092.

Yusof, N.N., Ghoshal, S.K., Arifin, R., Awang, A., Tewari, H.S. & Hamzah, K. 2018. Self-cleaning and Spectral Attributes of Erbium-doped Sodium-zinc Tellurite Glass: Role of Titania Nanoparticles. *Journal of Non-Crystalline Solids*, 481, 225-238.

Yusoff, N.M. & Sahar, M.R. 2015. Effect of Silver Nanoparticles Incorporated with Samarium-doped Magnesium Tellurite Glasses. *Physica B: Condensed Matter*, 456, 191-196.

Zhang, S., Zhang, L., Liu, K., Liu, M., Yin, Y. & Gao, C. 2018. Digestive Ripening in the Formation of Monodisperse Silver Nanospheres. *Materials Chemistry Frontiers*, 2, 1328-1333.

Zhu, M., Lei, B., Ren, F., Chen, P., Shen, Y., Guan, B., Du, Y., Li, T. & Liu, M. 2014. Branched Au Nanostructures Enriched with a Uniform Facet: Facile Synthesis and Catalytic Performances. *Scientific Reports*, 4, 5259.

Żmojda, J., Kochanowicz, M. & Dorosz, D. 2014. Low-phonon Tellurite Glass Co-doped with Tm^{3+}/Ho^{3+} Ions for Optical Fiber Technology. *Photonics Letters of Poland*, 6(2), 56-58.

UMS
UNIVERSITI MALAYSIA SABAH

Zschornack, G. H. 2007a. Physical Fundamentals. *In Handbook of X-ray Data*, pp. 24-35. Berlin, Germany: Springer.

Zschornack, G. H. 2007b. X-ray Emission Lines and Atomic Level Characteristics. *In Handbook of X-ray Data,* pp. 179-486. Berlin, Germany: Springer.

Zschornack, G. H. 2007c. Periodic Table of the Elements. *In Handbook of X-ray Data*. Berlin, Germany: Springer.

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UNIVERSITI MALAYSIA SABAH