## **ORIGINAL ARTICLE**





# A New *Epi*-neoverrucosane-type Diterpenoid from the Liverwort *Pleurozia subinflata* in Borneo

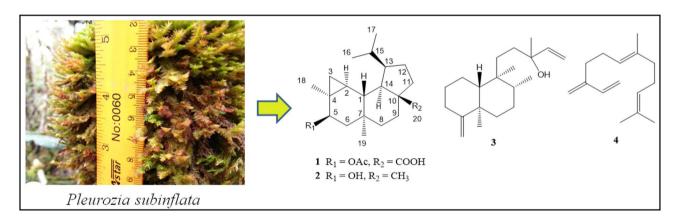
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#### **Abstract**

New bioactive 13-epi-neoverrucosane diterpenoid,  $5\beta$ -acetoxy-13-epi-neoverrucosanic acid (1) along with three known secondary metabolites, 13-epi-neoverrucosan- $5\beta$ -ol (2), chelodane (3) and (E)- $\beta$ -farnesene (4) were isolated from the MeOH extract of east Malaysia's liverwort *Pleurozia subinflata*. The chemical structure of new compound was elucidated by the analyses of its spectroscopic data (FTIR, NMR and HR-ESI-MS). These epi-neoverrucosane-type compounds seem to be notable chemosystematic markers for *P. subinflata* in Borneo. Compound 3 was widespread in marine sponges however this is the first record for 3 to be found in liverwort. These metabolites were tested for their antifungal potentials against selected fungi from the marine environment. Compound 1 exhibited effective antifungal activity against *Lagenidium thermophilum*.

#### **Graphic Abstract**



Keywords Epi-neoverrucosane · Diterpenoid · Pleurozia subinflata · Liverwort · Borneo

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## 1 Introduction

Liverworts are the largest group of pioneering land plants which arose during the adaptation of plants from marine to terrestrial environment [1]. They produced terpenoids and/ or aromatic compounds as their major lipophilic constituents [2–5]. Many types of sesquiterpenoids from liverworts are the enantiomer to those metabolites from higher plants [6]. While, diterpenoids such as clerodanes, dolabellanes, fusicoccanes, kauranes, labdanes, pimaranes and others were found in numerous liverworts [6]. Recently several bioactive



cyathane diterpenoids were discovered [7, 8]. Cyathane is precursor structure for biosynthesis of the verrucosane-type diterpenoid, a fused 3,6,6,5-tetracyclic carbon skeleton [9]. First verrucosane diterpenoid was isolated from *Mylia verrucosa* [10]. Later, some analogs such as neoverrucosane-, homoverrucosane-, *epi*-neoverrucosane- and *epi*-homoverrucosane-type were reported [11–13]. The latest *epi*-neoverrucosane analog was reported in 2013 [14]. Hereby, we report yet another new *epi*-neoverrucosane diterpenoid,  $5\beta$ -acetoxy-13-*epi*-neoverrucosanic acid (1) was isolated, along with three known secondary metabolites, 13-*epi*-neoverrucosan- $5\beta$ -ol (2), chelodane (3) and (*E*)- $\beta$ -farnesene (4) from east Malaysia's liverwort *Pleurozia subinflata* (Fig. 1).

Besides, liverworts have long been used as traditional medicine by the indigenous people in some parts of China. In the past half-century, Prof. Yoshinori Asakawa (Tokushima Bunri University, Japan) has begun to study the chemical composition of liverworts collected from Asia, Europe and South America, many of which have reported to have diverse chemical structures and exhibited numerous biological activities [2, 6]. Thus, research focusing on the biological activity of liverwort and industrial use are significant. Our research examined the antifungal effects of the four isolated compounds against selected fungi separated marine organisms.

#### 2 Results, Discussion and Conclusion

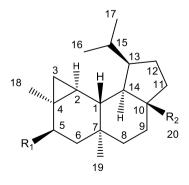
Compound **1** was isolated as colorless oil and analyzed for the molecular formula  $C_{22}H_{34}O_4$  by HR-ESI-MS [M – H]<sup>-</sup> ion at m/z 361.2391. The <sup>13</sup>C and <sup>1</sup>H NMR data (Table 1) indicated the presence of an isopropyl unit at  $\delta_{\rm C}$  31.1, 23.2 and 22.3;  $\delta_{\rm H}$  1.51–1.55, 1.00 and 0.78, one carboxylic carbon at  $\delta_{\rm C}$  183.0, an acetoxy unit at  $\delta_{\rm C}$  171.5, 21.5;  $\delta_{\rm H}$  2.04, one oxygenated methine at  $\delta_{\rm C}$  75.1;  $\delta_{\rm H}$  5.28, two tertiary methyls at  $\delta_{\rm C}$  25.5 and 16.4;  $\delta_{\rm H}$  1.12 and 0.87,

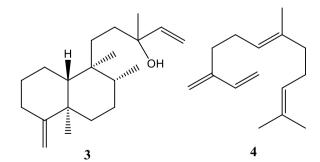
**Table 1**  $^{1}$ H and  $^{13}$ C NMR (600 and 150 MHz) data of **1** (in CDCl<sub>3</sub>,  $\delta$  in ppm, J in Hz).

No	$\delta_{ m C}$	$\delta_{ m H}$	
1	44.1	2.20 (1H, dd, <i>J</i> =4.3, 13.1)	
2	26.7	0.65 (1H, ddd, J=4.1, 4.1, 8.3)	
3	21.8	0.69 (1H, dd, J=4.3, 8.3)	
		0.56 (1H, dd, $J=4.3$ , $4.3$ )	
4	20.1		
5	75.1	5.28 (1H, dd, J=7.6, 10.7)	
6	42.3	1.76 (1H, dd, $J$ =7.6, 12.5)	
		0.84 (1H, m)	
7	36.8		
8	34.9	1.51–1.55 (1H, m)	
		1.09–1.11 (1H, m)	
9	32.3	1.81–1.84 (1H, m)	
		1.59–1.62 (1H, m)	
10	52.0		
11	36.3	2.45  (1H, ddd,  J = 1.4, 8.9, 11.6)	
		1.21–1.23 (1H, m)	
12	27.5	1.81–1.84 (1H, m)	
		1.51–1.55 (1H, m)	
13	45.3	1.88 (1H, m)	
14	51.7	1.67 (1H, dd, $J$ =7.6, 13.1)	
15	31.1	1.51–1.55 (1H, m)	
16	22.3	0.78 (3H, d, J = 6.9)	
17	23.2	1.00 (3H, d, J=6.9)	
18	25.5	1.12 (3H, s)	
19	16.4	0.87 (3H, s)	
20	183.0		
5-OAc	171.5		
	21.5	2.04 (3H, s)	

six methylenes, four methines, and three quaternary carbons which corresponding well to HSQC spectrum. Based on these findings, six degrees of unsaturation was

Fig. 1 Chemical structures of 1–4







calculated from HR-ESI-MS, and it attributed to two carbonyl units and one tetracyclic ring for 1.

The  ${}^{1}\text{H}{}^{-1}\text{H}$  COSY experiment revealed the spin systems as depicted by the bold lines in Fig. 2. In the HMBC spectrum, the three-bond correlations of  $H_3$ –C(16) and  $H_3$ –C(17) to the opposite carbons C(17) and C(16), and to C(13) and C(15), allowed the placement of isopropyl unit at C(13) which was further confirmed by  ${}^{1}\text{H}{}^{-1}\text{H}$  COSY. The acetoxy unit at C(5) was confirmed by HMBC correlations between H–C(5) to 5-OAc. The downfield shifted of  ${}^{13}\text{C}$  and  ${}^{1}\text{H}$  NMR at C(5) further supported this deduction. The HMBC correlations of  ${}^{1}\text{H}{}^{-1}\text{C}$  C(11) to C(20); and H–C(14) to C(20) suggested the carboxylic carbon at C(10). These findings together with HMBC correlations of  ${}^{1}\text{H}{}^{-1}\text{C}$  C(18) to C(2), C(3), C(4) and C(5); and  ${}^{1}\text{H}{}^{-1}\text{C}$  C(19) to C(1), C(6), C(7) and C(8) permitted establishment for the planar structure of 1 (Fig. 2).

The relative stereochemistry of 1 was deduced from the NOESY correlations (Fig. 2) and comparison of its chemical shift, coupling constants and NOE correlations with those of known analogs [10-14]. The NOE correlations of H–C(5) to H<sub>2</sub>–C(6 $\alpha$ ) ( $\delta$ <sub>H</sub> 1.76), H<sub>3</sub>–C(18) and H<sub>3</sub>–C(19); and H–C(14) to  $H_3$ –C(19) have suggested H–C(5), H–C(14),  $H_3$ –C(18) and  $H_3$ –C(19) on  $\alpha$  relative configuration. While, NOE correlations of H–C(1) to  $H_2$ –C(3 $\beta$ ) ( $\delta_H$  0.56) and  $H_2$ – $C(6\beta)$  ( $\delta_H$  0.84) showed H–C(1) must be on another face, β relative configuration. The earlier NOE cross peak of H–C(1) to H<sub>2</sub>–C(3) ( $\delta_{\rm H}$  0.56) has led to the assignment of H<sub>2</sub>–C(3) ( $\delta_{\rm H}$  0.56) on  $\beta$  configuration, therefore H<sub>2</sub>–C(3)  $(\delta_{\rm H}\,0.69)$  must be on  $\alpha$  configuration. With this finding, the configuration at H–C(2) can be assigned based on vicinal coupling constants of cyclopropane unit between H–C(2) and  $H_2$ –C(3 $\alpha$ ) ( ${}^3J_{2-3\alpha}$  = 8.3 Hz) and between H–C(2) and  $H_2$ –C(3 $\beta$ ) ( ${}^3J_{2-3\beta}$ =4.3 Hz). These coupling constant values suggested H–C(2) has a *cis* relationship with  $H_2$ –C(3 $\alpha$ ) within the cyclopropane unit, therefore  $\alpha$  configuration was assigned for H–C(2). While, the carboxyl unit at C(10) was assigned on β configuration due to a trans-fused at C/D ring junction. Thus, the relative configurations of  $1R^*$ ,  $2S^*$ ,  $4S^*$ ,  $5R^*$ ,  $7S^*$ ,  $10S^*$  and  $14R^*$  were determined as identical to those of known analogs of neoverrucosane and epi-neoverrucosane [10–14]. To distinguish *epi*-neoverrucosane from neoverrucosane, the NOE correlations of H-C(1) to H-C(15) and  $H_3-C(17)$ ; and H-C(13) to H-C(14), showed 13-isopropyl unit to H–C(14) has a trans configuration, suggested a epi-neoverrucosane. Furthermore, similar NOE correlations of H-C(1) to H-C(15); and H-C(13) to H-C(14) were observed in 12-acetoxy-13-epi-neoverrucosann-5-one [14]. On the contrary, these NOE were not observed in neoverrucosane-type, neoverrucosan-5β,9β-diol, instead H–C(13) to H–C(20) was detected [15]. Thus, the structure 1 was established without confusion. The configuration of isopropyl unit at C(13) generated during formation of tricyclic system (Fig. 3) determined the biosynthesis of neoverrucosane or epi-neoverrucosane [9]. To the best of our knowledge, compound 1 was considered as first discovery of 13-epi-neoverrucosane that containing a carboxyl moiety or even among related skeletons such as verrucosane and neoverrucosane. The methyl at C(20) of 1 might have followed a three-step oxidation, via a hydroxyl and carbonyl, to the corresponding carboxylic acid [16], as shown in the purposed biosynthetic pathway (Fig. 3).

The known compounds were identified as 13-epineoverrucosan- $5\beta$ -ol (2) [11], chelodane (3) [17], and (E)β-farnesene (4) [18], after compared its spectroscopic data with published literatures. The tetracyclic diterpenes are relatively rare in nature, and mainly found in the species of Plagiochila, Jamesoniella and Fossombronia [6]. However, we found *epi*-neoverrucosane-type diterpene derivatives (1 and 2) from east Malaysia's liverwort, *Pleurozia subinflata*. These secondary metabolites seem to be the good chemosystematic markers for *P. subinflata* in Borneo. Compound 3 was widespread in marine sponges such as Chelonaplysilla erecta, Raspailia sp. and even in Sigmosceptrella sp. [17, 19, 20]. However, this is the first record for 3 found in liverwort. Compound 4 was the most common farnesane-type sesquiterpene in liverworts. It was distributed throughout more than 20 Jungermanniales and Pleuroziales species including *Pleurozia* [6].

Fig. 2 <sup>1</sup>H-<sup>1</sup>H COSY, key HMBC and NOE correlations of 1

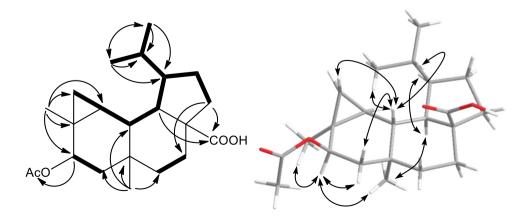




Fig. 3 Proposed biosynthetic pathway of 1

Table 2 The MIC (μg/mL) of 1–4 against six fungal strains

Strains	MIC (µg/mL)				
	1	2	3	4	
L. thermophilum	12.5	100.0	100.0	> 100.0	
H. sabahensis	50.0	> 100.0	> 100.0	> 100.0	
F. moniliforme	>100.0	> 100.0	> 100.0	> 100.0	
F. oxysporum	>100.0	> 100.0	>100.0	> 100.0	
F. solani	>100.0	> 100.0	> 100.0	> 100.0	
O. humicola	>100.0	> 100.0	> 100.0	> 100.0	

Positive control: itraconazole MIC 2.0 µg mL<sup>-1</sup>

Compounds 1–4 were evaluated its biological potentials against fungal strains isolated from the Bornean ocean, Lagenidium thermophilum IPMB 1401, Haliphthoros sabahensis IPMB 1402, Fusarium moniliforme NJM 8995, Fusarium oxysporum NJM 0179, Fusarium solani NJM 8996 and Ochroconis humicola NJM 1503 (Table 2). The minimum inhibition concentration (MIC) values of compound 1 against L. thermophilum and H. sabahensis were 12.5 and 50 μg/mL, respectively. While, compounds 2 and 3 showed MIC values of 100 μg/mL against L. thermophilum. Compound 4 was inactive (> 100 μg/mL) against the tested fungi.

In conclusion, this is the first time ever since 2013 of the last epi-neoverrucosane being discovered from nature. A new 13-epi-neoverrucosane diterpenoid, 5 $\beta$ -acetoxy-13-epi-neoverrucosanic acid (1) along with three known secondary metabolites, 13-epi-neoverrucosan-5 $\beta$ -ol (2), chelodane (3) and (E)- $\beta$ -farnesene (4) were found in east Malaysia's liverwort Pleurozia subinflata. Compound

1 exhibited effective antifungal activity (MIC values of 12.5 μg/mL) against *Lagenidium thermophilum*.

## 3 Experimental Section

## 3.1 General Experimental Procedures

Optical rotation was taken on the automatic polarimeter (AUTOPOL IV automatic polarimeter) in chloroform solutions at 28 °C. IR spectrum was recorded on the FTIR spectroscopy (Perkin Elmer). NMR spectra were recorded on the 600 MHz FT-NMR (Jeol) using deuterated chloroform (CDCl<sub>3</sub>) with tetramethylsilane (TMS) as the internal standard. MS spectra were obtained using LC-ESI-IT-TOF-MS (Shimadzu). For preparative TLC, Merck Kieselgel 60 F<sub>254</sub> was used and Kieselgel 60 was used for column chromatography. Purification was performed using high performance liquid chromatography (LC-10 AT, Shimadzu) equipped with UV detector.

#### 3.2 Biological Materials

Specimens of *P. subinflata* (*M. Suleiman & S.-Y. Ng* 5946) were collected from Mount Trus Madi (5° 33′ 13.1″ N, 116° 30′ 41.9″ E), Sabah, Malaysia in August 2015. The specimens were identified based on external morphology by the fifth author. A voucher specimen (BORHB0026) is deposited in the BORNEENSIS Herbarium at Institute for Tropical Biology and Conservation (ITBC), Univeristi Malaysia Sabah (UMS).



#### 3.3 Extraction and Isolation

The air-dried liverwort specimens (42 g) were extracted using 100% methanol (MeOH) (1.0 L × 3 each for two days). The crude extract was partitioned between distilled water (150 mL) and ethyl acetate (EtOAc) (50 mL  $\times$  3). After removal of the organic solvent, the EtOAc fraction (500 mg) was chromatographed on a Si gel column using hexane (Hex) and EtOAc system as eluent with increasing polarity (Hex/EtOAc: 9:1, 8:2, 7:3, 5:5, 100% EtOAc) to yield five fractions, 1–5. Fraction 2 (76 mg) was subjected to repeated preparative TLC with toluene to yield 2 (8.8 mg), 3 (7.4 mg) and 4 (15.4 mg). Fraction 3 (60 mg) was subjected to repeated preparative TLC with hexane/EtOAc: 7:3, and the resulted sub-fraction was further purified by semi-preparative high performance liquid chromatography (HPLC) to yield 1 (12.8 mg). The isolation was operated using a reverse phase  $C_{18}$  column (5 µm, 10 mm × 250 mm) measured at UV wavelength of 210 nm under gradient elution with the following conditions: 40–100% acetonitrile (MeCN)/H<sub>2</sub>O.

#### 3.3.1 5β-Acetoxy-13-*Epi*-neoverrucosanic Acid (1)

Colorless oil;  $[\alpha]_D^{28.0} + 67.8$  (c = 0.5, CHCl<sub>3</sub>); IR  $\nu_{max}$  3488, 3060, 1735 and 1712 cm<sup>-1</sup>;  ${}^{1}$ H NMR (CDCl<sub>3</sub>, 600 MHz)  $\delta$ 5.28 (1H, dd, J = 7.6, 10.7 Hz, H-5), 2.45 (1H, ddd, J = 1.4, 8.9, 11.6 Hz, H-11 $\alpha$ ), 2.20 (1H, dd, J=4.3, 13.1 Hz, H-1), 2.04 (3H, s, OAc), 1.88 (1H, m, H-13), 1.81-1.84  $(1H, m, H-12\alpha), 1.81-1.84$   $(1H, m, H-9\alpha), 1.76$  (1H, dd,J = 7.6, 12.5 Hz, H-6 $\alpha$ ), 1.67 (1H, dd, J = 7.6, 13.1 Hz, H-14), 1.59–1.62 (1H, m, H-9 $\beta$ ), 1.51–1.55 (1H, m, H-8 $\alpha$ ), 1.51-1.55 (1H, m, H-12 $\beta$ ), 1.51-1.55 (1H, m, H-15), 1.21–1.23 (1H, m, H-11β), 1.12 (3H, s, H-18), 1.09–1.11  $(1H, m, H-8\beta), 1.00 (3H, d, J=6.9 Hz, H-17), 0.87 (3H, s,$ H-19), 0.84 (1H, m, H-6 $\beta$ ), 0.78 (3H, d, J=6.9 Hz, H-16),  $0.69 (1H, dd, J=4.3, 8.3 Hz, H-3\alpha), 0.65 (1H, ddd, J=4.1,$ 4.1, 8.3, H-2), 0.56 (1H, dd, J = 4.3, 4.3 Hz, H-3 $\beta$ ); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz)  $\delta$  183.0 (C, C-20), 171.5 (C, OAc), 75.1 (CH, C-5), 52.0 (C, C-10), 51.7 (CH, C-14), 45.3 (CH, C-13), 44.1 (CH, C-1), 42.3 (CH<sub>2</sub>, C-6), 36.8 (C, C-7), 36.3 (CH<sub>2</sub>, C-11), 34.9 (CH<sub>2</sub>, C-8), 32.3 (CH<sub>2</sub>, C-9), 31.1 (CH, C-15), 27.5 (CH<sub>2</sub>, C-12), 26.7 (CH, C-2), 25.5 (CH<sub>3</sub>, C-18), 23.2 (CH<sub>3</sub>, C-17), 22.3 (CH<sub>3</sub>, C-16), 21.8 (CH<sub>2</sub>, C-3), 21.5 (CH<sub>3</sub>, OAc), 20.1 (C, C-4), 16.4 (CH<sub>3</sub>, C-19); negative ion HRESIMS: m/z 361.2391 (calcd for  $C_{22}H_{33}O_4$  [M–H]<sup>-</sup>, 361.2384).

## 3.4 Antifungal Assay

The minimum inhibitory concentration (MIC) of the fungistatic on hyphae was performed by incorporating the pure compound solutions (12.5, 25.0, 50.0, 100.0 µg/mL) onto PYGS agar in a petri dish followed by inoculation of six

tested fungal strains [21–23]. The MIC was determined visually as the lowest concentration showing no hyphal growth after they were incubated at 25 °C for 7 days.

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#### **Compliance with Ethical Standards**

**Conflict of interest** The authors declare no conflict of interest.

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## References

- L. Hui, R.X. Xu, X.S. Zhang, T.T. Zhu, H.X. Lou, A.X. Cheng, Phytochemistry 159, 190–198 (2019)
- 2. Y. Asakawa, A. Ludwiczuk, J. Nat. Prod. 81, 641–660 (2018)
- S.Y. Ng, T. Kamada, C.S. Vairappan, Rec. Nat. Prod. 11, 508– 513 (2017)
- S.Y. Ng, T. Kamada, M. Suleiman, C.S. Vairappan, Nat. Prod. Res. 32, 1832–1837 (2018)
- S.Y. Ng, T. Kamada, C.S. Phan, M. Suleiman, C.S. Vairappan, Heterocycles 96, 1958–1965 (2018)
- Y. Asakawa, A. Ludwiczuk, F. Nagashima, in Chemical Constituents of Bryophytes, ed. by A.D. Kinghorn. Progress in the Chemistry of Organic Natural Products, vol 95 (Springer, Vienna, 2013), p. 796
- J. Wei, Y. Cheng, W.H. Guo, D.C. Wang, Q. Zhang, D. Li, J. Rong, J.M. Gao, Sci. Rep. 7, 8883–8896 (2017)
- J. Wei, W.H. Guo, C.Y. Cao, R.W. Kou, Y.Z. Xu, M. Górecki, L.D. Bari, G. Pescitelli, J.M. Gao, Sci. Rep. 8, 2175–2189 (2018)
- 9. P.M. Dewick, Nat. Prod. Rep. 19, 181–222 (2002)
- A. Matsuo, H. Nozaki, M. Nakayama, S. Hayashi, J. Chem. Soc. Chem. Commun. 5, 198–200 (1978)
- C. Grammes, G. Burkhardt, M. Veith, V. Hugh, H. Becker, Phytochemistry 44, 1495–1502 (1997)
- H.J. Liu, C.L. Wu, H. Becker, J. Zapp, Phytochemistry 53, 845–849 (2000)



- H. Shimogawa, T. Teruya, K. Suenaga, H. Kigoshi, Bull. Chem. Soc. Jpn. 78, 1345–1347 (2005)
- A.J. Singh, J.D. Dattelbaum, J.J. Field, Z. Smart, E.F. Woolly, J.M. Barber, R. Heathcott, J.H. Miller, P.T. Northcote, Org. Biomol. Chem. 11, 8041–8051 (2013)
- A. Spyere, D.C. Rowley, P.R. Jensen, W. Fenical, J. Nat. Prod. 66, 818–822 (2003)
- 16. A. Banerjee, B. Hamberger, Phytochem. Rev. 17, 81–111 (2018)
- 17. A. Rudi, Y. Kashman, J. Nat. Prod. 55, 1408–1414 (1992)
- Y. Asakawa, M. Toyota, H. Tanaka, T. Hashimoto, D. Joulain, J. Hattori Bot. Lab. 78, 183–188 (1995)
- T. Yosief, A. Rudi, Z. Stein, I. Goldberg, G.M.D. Gravalos, M. Schleyer, Y. Kashman, Tetrahedron Lett. 39, 3323–3326 (1998)
- J.M. Ryan, Ph.D Dissertation, Victoria University of Wellington, 2007
- T. Kamada, C.S. Phan, T. Hamada, K. Hatai, C.S. Vairappan, Nat. Prod. Commun. 13, 17–19 (2018)
- T. Kamada, J. Kulip, K. Tani, C.S. Phan, K. Hatai, C.S. Vairappan, Rec. Nat. Prod. 12, 317–322 (2018)
- K.J. Shamsudin, C.S. Phan, J. Kulip, K. Hatai, C.S. Vairappan, T. Kamada, J. Asian Nat. Prod. Res. 21, 435–441 (2019)

